

1. $\text{Fe(s)} + \text{H}_2\text{SO}_4\text{(aq)} \rightarrow \text{Fe}_2(\text{SO}_4)_3\text{(aq)} + \text{H}_2\text{O(l)}$
2. $\text{NaC}_2\text{H}_3\text{O}_2\text{(aq)} + \text{H}_2\text{CO}_3\text{(aq)} \rightarrow \text{Na}_2\text{CO}_3\text{(aq)} + \text{H}_2\text{O(l)} + \text{C}_2\text{H}_5\text{O(l)}$
3. $\text{(NH}_4)_3\text{PO}_4\text{(aq)} + \text{Pb(NO}_3)_4\text{(aq)} \rightarrow \text{Pb}_3(\text{PO}_4)_2\text{(s)} + \text{NH}_4\text{NO}_3\text{(aq)}$
4. $\text{C}_4\text{H}_{10}\text{(g)} + \text{O}_2\text{(g)} \rightarrow \text{CO}_2\text{(g)} + \text{H}_2\text{O(g)}$
5. $\text{H}_3\text{PO}_4\text{(aq)} + \text{HCl(aq)} \rightarrow \text{PCl}_5\text{(s)} + \text{H}_2\text{O(l)}$
6. $\text{Hg(OH)}_2\text{(s)} + \text{H}_3\text{AsO}_4\text{(aq)} \rightarrow \text{Hg}_3(\text{AsO}_4)_2\text{(s)} + \text{H}_2\text{O(l)}$
7. $\text{HClO}_4\text{(aq)} + \text{P}_2\text{O}_{10}\text{(s)} \rightarrow \text{H}_2\text{PO}_4\text{(aq)} + \text{ClO}_4\text{(aq)} + \text{P}_2\text{O}_5\text{(s)}$

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