

Multiple Choice:

Selecting the alternative that constitutes the statement or answers the question. What is most commonly observed with water?

1. How many atoms of hydrogen (protons) is a molecule of liquid water made of?
 a. 1
 b. 2
 c. 3
 d. 4

2. At seawater $pH = 8.1$, what is the approximate percentage of bicarbonate?
 a. 99
 b. 99.9
 c. 99.99
 d. 99.999

3. The uncharged state occurs in the seawater at:
 a. pH of the oceans
 b. pH of the atmosphere
 c. pH of the hydrosphere
 d. pH of the lithosphere

4. Full seawater $pH = 8.1$, $10^{-8.1} = 7.9 \times 10^{-9}$. How many moles of H^+ are there per mole of bicarbonate combined at that pH coefficient?
 a. 10 fold
 b. 100 fold
 c. 1000 fold
 d. 10000 fold

5. What is the state of the water molecules in the uncharged products in seawater?
 a. charged molecules
 b. uncharged molecules
 c. charged ions
 d. uncharged ions

6. What is the basic neutral distribution?
 a. 100% acid
 b. 100% base
 c. 100% neutral
 d. 100% mixed

7. A compound's empirical formula is CH_2 , at that ratio, what is its grams of hydrogen relative to the carbon?
 a. 1/12
 b. 1/6
 c. 1/3
 d. 1/2

8. Full seawater $pH = 8.1$, $10^{-8.1} = 7.9 \times 10^{-9}$. How many grams of a strong protonated acid (H) per mole bicarbonate?
 a. 1000
 b. 10000
 c. 100000
 d. 1000000

9. At seawater $pH = 8.1$, $10^{-8.1} = 7.9 \times 10^{-9}$, what is the approximate volume in million fold?
 a. 10
 b. 100
 c. 1000
 d. 10000

10. How many moles of protons (H) are there per mole of bicarbonate?
 a. 1000
 b. 10000
 c. 100000
 d. 1000000

11. How many moles of H^+ are there per mole of bicarbonate?
 a. 1000
 b. 10000
 c. 100000
 d. 1000000

Multiple Choice Stoichiometry Test

Angela Woodward Spangenberg

A red circular graphic with a gradient, appearing as a partial circle or a thick arc, located to the right of the author's name.

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