

Chemistry Study Guide

Periodic Table of the Elements																		
H	Li	Be																
Na	Mg																	
K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr	
Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe	
Cs	Ba	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn		
Fr	Ra	Rf	Db	Sg	Bh	Hs	Mt	Ds	Rg	Cn	Uut	Fl	Uup	Lv	Uus	Uuo		
		La	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu		
		Ac	Th	Pa	U	Ng	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr		

Objectives:

To understand the structure of an atom, and the nature + properties of subatomic particles.



Matter:

Anything that has mass and occupies space.

Examples: air, gases, liquids, solidz!

Atoms:

Atoms are the smallest, indivisible particle.

- Outer part → shell/orbit/energy level → carries the electronz (e-)
- Inner part → nucleus → houses the protonz (p+) and neutronz (n)
- Outer & inner part → together these are the subatomic particles.

Subatomic Particle Table:

Particle	Symbol	Mass (amu)	Charge	Location
Proton	p ⁺	1	+1	Nucleus
Neutron	n	1	0	Nucleus
Electron	e ⁻	-0	-1	Orbit/Shell

Section 2 nuclear Chemistry Study Guide Key

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General Chemistry Study Guide Sixth Edition Darrell D. Ebbing, 1999 **Guide**

to U.S. Government Publications Donna Batton, 2007-08 This comprehensive yet concise annual annotated reference source catalogs the important series, periodicals and reference tools published by U.S. government agencies. Over the years the index section of the Guide to U.S. Government Publications has expanded to more than 40,000 entries. Agencies and titles are indexed followed by a keyword title index for quick and easy referencing. No other single resource provides historical and current information on U.S. government publications in one place.

Chemistry Charles E. Mortimer, 1983 Good. No Highlights. No Markup. All pages are intact. Slight shelfwear may have the corners slightly dented. May have slight color changes. Slightly damaged spine.

Chemistry George M. Bodner, Harry L. Pardue, 1989-01-23 A modern experimental approach to first year chemistry. This unique introductory account employs experimental observations to construct the principles of general chemistry. An early introduction to observable descriptive chemistry lays the basis for the well developed exposition that follows.

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Section 2nuclear Chemistry Study Guide Key Book Review: Unveiling the Magic of Language

In a digital era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Section 2nuclear Chemistry Study Guide Key**," written by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound affect our existence. Throughout this critique, we shall delve in to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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