

Name _____ Group _____

Chemistry 2nd Semester
Review 4th Period exam

1.- Identify as organic or inorganic

- a) KCl b) C₂H₆O c) CaCl₂ d) CO₂ e) NaCN f) CH₄

2.- Identify the following properties as organic or inorganic

- a) soluble in water b) burns in air c) low boiling point d) contain K
e) form ionic bond f) form isomers g) high boiling point h) can be used as fuel

3.- True or false

- a) Carbon atoms can bond with other carbon atoms any number of times
b) Carbon atomic number is 4
c) Carbon forms 4 covalent bonds
d) Carbon can bond with many elements, metals and nonmetals

4.- Identify the following as alkane, alkene or alkyne

- a) $\text{HC}\equiv\text{CCH}_2\text{CH}_2\text{CH}_3$ b) $\text{CH}_3-\text{CH}_2-\underset{\text{CH}_3}{\text{CH}}-\text{CH}_2-\text{CH}_2-\text{CH}_3$ c) CH₂=CH-CH₃

5.- Identify as saturated or unsaturated

- a) $\text{CH}_3-\underset{\text{CH}_3}{\text{CH}}-\underset{\text{CH}_3}{\text{C}}=\text{CH}_2$ b) $\text{H}_2\text{C}-\underset{\text{CH}_3}{\text{CH}}-\text{CH}_3$

6.- Give the name of the following alkyl groups

- a) -CH₃ b) -CH₂-CH₃ c) -CH₂-CH₂-CH₃ d) CH₃-CH-CH₃

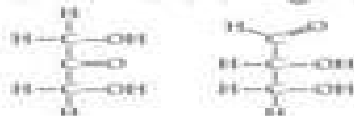
7.- What are aliphatic hydrocarbons.

8.- Identify as hydrocarbon or derivate of hydrocarbon

- a) CH₃-Cl b) CH₃-CH₂-CH₃ c) CH₃-COOH d) CH₃-CH=CH-CH₃
e) CH₃-O-CH₃ f) C₆H₆ g) CH₃-CH=O

9.- What are isomers?

10.- Do the following molecules represent isomers? Explain.



11.- Circle the functional group and identify the class or family.

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