

Multiple Choice

Selecting the alternative that constitutes the statement or answers the question. What is most commonly observed with water?

1. How many moles of hydroxide groups are a molecule of typical silicate tetrahedron?
 a. 1
 b. 2
 c. 3
 d. 4

2. For the reaction $\text{Si}_2\text{O}_2 + 2\text{H}_2\text{O} \rightarrow 2\text{SiOH}_2$, what factor determines whether a reaction?
 a. ΔG
 b. ΔH
 c. ΔS
 d. ΔG°

3. The unhydrated silicate anion in the fused sil.
 a. SiO_4^{4-}
 b. SiO_3^{2-}
 c. SiO_2
 d. SiO_4^{2-}

4. For the reaction $\text{Si}_2\text{O}_2 + 2\text{H}_2\text{O} \rightarrow 2\text{SiOH}_2$, ΔG° for the reaction is 0 kJ/mol. The reaction is spontaneous?
 a. Yes
 b. No
 c. Not sure
 d. Cannot tell

5. What is the study of the relationships between mineral products in natural systems?
 a. Mineralogy
 b. Petrology
 c. Geochemistry
 d. Mineralogy

6. What is the basic mineral structure?
 a. SiO_4^{4-}
 b. SiO_3^{2-}
 c. SiO_2
 d. SiO_4^{2-}

7. A mineral's empirical formula is Si_2O_2 , at this ratio, what is the charge on the silicate tetrahedron?
 a. SiO_4^{4-}
 b. SiO_3^{2-}
 c. SiO_2
 d. SiO_4^{2-}

8. For the reaction $\text{Si}_2\text{O}_2 + 2\text{H}_2\text{O} \rightarrow 2\text{SiOH}_2$, ΔG° for the reaction is 0 kJ/mol. The reaction is spontaneous?
 a. Yes
 b. No
 c. Not sure
 d. Cannot tell

9. For the reaction $\text{Si}_2\text{O}_2 + 2\text{H}_2\text{O} \rightarrow 2\text{SiOH}_2$, ΔG° for the reaction is 0 kJ/mol. The reaction is spontaneous?
 a. Yes
 b. No
 c. Not sure
 d. Cannot tell

10. What is the basic mineral structure?
 a. SiO_4^{4-}
 b. SiO_3^{2-}
 c. SiO_2
 d. SiO_4^{2-}

11. The unhydrated silicate anion in the fused sil.
 a. SiO_4^{4-}
 b. SiO_3^{2-}
 c. SiO_2
 d. SiO_4^{2-}

Multiple Choice Stoichiometry Test

Mary Millhollon, Richard H. Langley



Multiple Choice Stoichiometry Test:

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