

## TEST YOUR SELF

### TEST YOURSELF 2.1 (THEORIES AND EXPERIMENTS RELATED TO STRUCTURE OF ATOMS)

i) Do you know any element having no neutrons in its atoms?

Ans: Yes the isotope of hydrogen (Protium,  ${}^1_1\text{H}$ ) has no neutron. It has one proton in its nucleus and one electron revolving around it.

ii) Who discovered an electron, a proton and a neutron?

Ans: a. In 1886, Goldstein discovered positively charged particles called protons.

b. In 1897, J.J. Thomson discovered the negatively charged particles called electrons.

c. In 1932, James Chadwick discovered the neutral particles called neutrons.

iii) How does electron differ from a neutron?

Ans: Difference between electron and neutron.

Electron	Neutron
i. Electron is the negatively charged particle.	i. Neutron is the neutral particle.
ii. Electron revolves around the nucleus.	ii. Neutron is present in the nucleus.
iii. Mass of electron is $5.486 \times 10^{-4}$ g or $9.106 \times 10^{-28}$ g.	iii. Mass of neutron is 1.0087 amu or $1.674 \times 10^{-24}$ g.

iv) Explain how anode rays are formed from the gas taken in the discharge tube?

Ans: Anode rays do not originate from the anode. In fact these rays are produced when the cathode rays or electrons collide with the residual gas molecules present in the discharge tube and ionize them as follows:



### TEST YOURSELF 2.2 RUTHERFORD'S ATOMIC MODEL AND BOHR'S ATOMIC MODEL

i) How was it proved that the whole mass of an atom is located at its centre?

Ans: Rutherford bombarded alpha particles on a 0.00004 cm thick gold foil. Almost all the particles passed through the foil undeflected, only few bounced back. The complete rebound of few alpha particles showed that the nucleus is an extremely small positively charged part. It is situated at the center of an atom and it carries nearly the whole mass of an atom.

ii) How was it shown that atomic nuclei are positively charged?

Ans: Alpha particles are helium nuclei ( $\text{He}^{2+}$ ) i.e. doubly positively charged particles. The deflection and rebounding of a few particles showed that atomic nuclei (centre of an atom) are positively charged.

iii) Name the particles which determine the mass of an atom.

Ans: Protons and neutrons present in the nucleus determine the mass of an atom. The sum of number of protons and neutrons is equal to mass of an atom.

iv) What is the classical theory of radiation? How does it differ from quantum theory?

Ans:

Classical theory of radiation	Quantum theory of radiation
According to classical theory of radiation electrons being the charged particles should release or emit energy continuously and they should ultimately fall into the nucleus.	According to quantum theory of radiation, energy is not emitted or absorbed continuously but it is emitted or absorbed in the form of small energy packets or bundles known as quantum or

# Quiz Yourself Chapter 8 Introductcion To Chemistry

**ACT**



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## **Table of Contents Quiz Yourself Chapter 8 Introductcion To Chemistry**

1. Understanding the eBook Quiz Yourself Chapter 8 Introductcion To Chemistry
  - The Rise of Digital Reading Quiz Yourself Chapter 8 Introductcion To Chemistry
  - Advantages of eBooks Over Traditional Books
2. Identifying Quiz Yourself Chapter 8 Introductcion To Chemistry
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Quiz Yourself Chapter 8 Introductcion To Chemistry
  - User-Friendly Interface
4. Exploring eBook Recommendations from Quiz Yourself Chapter 8 Introductcion To Chemistry
  - Personalized Recommendations
  - Quiz Yourself Chapter 8 Introductcion To Chemistry User Reviews and Ratings
  - Quiz Yourself Chapter 8 Introductcion To Chemistry and Bestseller Lists



5. Accessing Quiz Yourself Chapter 8 Introduction To Chemistry Free and Paid eBooks
  - Quiz Yourself Chapter 8 Introduction To Chemistry Public Domain eBooks
  - Quiz Yourself Chapter 8 Introduction To Chemistry eBook Subscription Services
  - Quiz Yourself Chapter 8 Introduction To Chemistry Budget-Friendly Options
6. Navigating Quiz Yourself Chapter 8 Introduction To Chemistry eBook Formats
  - ePub, PDF, MOBI, and More
  - Quiz Yourself Chapter 8 Introduction To Chemistry Compatibility with Devices
  - Quiz Yourself Chapter 8 Introduction To Chemistry Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Quiz Yourself Chapter 8 Introduction To Chemistry
  - Highlighting and Note-Taking Quiz Yourself Chapter 8 Introduction To Chemistry
  - Interactive Elements Quiz Yourself Chapter 8 Introduction To Chemistry
8. Staying Engaged with Quiz Yourself Chapter 8 Introduction To Chemistry
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Quiz Yourself Chapter 8 Introduction To Chemistry
9. Balancing eBooks and Physical Books Quiz Yourself Chapter 8 Introduction To Chemistry
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Quiz Yourself Chapter 8 Introduction To Chemistry
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Quiz Yourself Chapter 8 Introduction To Chemistry
  - Setting Reading Goals Quiz Yourself Chapter 8 Introduction To Chemistry
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Quiz Yourself Chapter 8 Introduction To Chemistry
  - Fact-Checking eBook Content of Quiz Yourself Chapter 8 Introduction To Chemistry
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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web remarks at rensselaer scholarship gala dinner program remarks shirley ann jackson ph d president rensselaer polytechnic institute good evening and welcome to our second east coast rensselaer scholarship gala which will be followed by an historic performance by the rensselaer orchestra at the world renowned carnegie hall

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web jun 19 2019 it gives me great pleasure as speaker to welcome you here tonight to mark a very special occasion for a very special member of the parliamentary press gallery tonight is ian s night and it is fitting that his friends and family colleagues and associates contacts and tipsters are gathered here in the banquet hall at parliament to honour him

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properly let us imagine that you are hosting a gathering

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web jun 13 2021 dinner 17 30 17 40 attendees are guided to their positions main hall from entrance 17 20 17 25 welcome speech 17 25 minute of silence optional 17 05 17 20 attendees arrive at the entrance thank you

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web 8 50 p m vvi speech to proceed with our ceremony i cordially invite the organizer of this event for his/her speech and officially officiate tonight's dinner please welcome speech thank you mr/ms for the warm opening speech just