

Multiple Choice

Select the alternative that correctly completes the statement or answers the question. What is the most common primary gas used?

1. How many moles of hydrogen gas are in a cylinder at 100 atm pressure, 20°C?
- ☐ 1
☒ 2
☐ 3
☐ 4

2. At constant $T_1 = 100^\circ\text{C}$, what is the molar volume of an ideal gas?
- ☒ 22.4
☐ 22.7
☐ 22.8
☐ 22.9

3. The universal gas constant is the same as:
- ☐ Boltzmann's constant
☐ gas constant
☒ both Boltzmann's and gas constant
☐ none of the above

4. For a constant $T_1 = 100^\circ\text{C}$, $n = 1.0$ mol, how many moles of gas are in a cylinder at 100 atm pressure, 20°C?
- ☐ 1.0 mol
☐ 2.0 mol
☐ 3.0 mol
☒ 4.0 mol

5. What is the value of the universal gas constant?
- ☐ 8.314 J/mol·K
☒ 8.314 J/mol·K
☐ 8.314 J/mol·K
☐ 8.314 J/mol·K

6. What is the value of the universal gas constant?
- ☐ 8.314 J/mol·K
☐ 8.314 J/mol·K
☒ 8.314 J/mol·K
☐ 8.314 J/mol·K

7. A cylinder of compressed hydrogen at 100 atm is at 20°C. How many moles of gas are in a cylinder at 100 atm pressure, 20°C?
- ☒ 2.0
☐ 2.0
☐ 2.0
☐ 2.0

8. For a constant $T_1 = 100^\circ\text{C}$, $n = 1.0$ mol, how many moles of gas are in a cylinder at 100 atm pressure, 20°C?
- ☒ 2.0
☐ 2.0
☐ 2.0
☐ 2.0

9. At constant $T_1 = 100^\circ\text{C}$, $n = 1.0$ mol, what is the molar volume of an ideal gas?
- ☐ 22.4
☐ 22.7
☒ 22.8
☐ 22.9

10. The universal gas constant is the same as:
- ☐ Boltzmann's constant
☐ gas constant
☒ both Boltzmann's and gas constant
☐ none of the above

11. The universal gas constant is the same as:
- ☐ Boltzmann's constant
☐ gas constant
☒ both Boltzmann's and gas constant
☐ none of the above

Multiple Choice Stoichiometry Test

Richard H. Langley



Multiple Choice Stoichiometry Test:

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