

Multiple Choice:

Select the alternative that represents the statement or answers the question. What is most commonly observed with water?

1. How many atoms of hydrogen (protons) is a molecule of liquid water made of?
 a. 1
 b. 2
 c. 3
 d. 4
2. At standard T_m (100°C) water has a hydrogen bonding network.
 a. Yes
 b. No
 c. Yes
 d. No
3. The hydrogen bond donor in the liquid state:
 a. Don't do anything
 b. Give electrons
 c. Give electronegativity
 d. Give electronegative
4. Pure water at T_m (100°C) is 100% hydrogen bonding network (hydrogen bonding network is 100% at 100°C).
 a. No
 b. Yes
 c. No
 d. Yes
5. What is the state of the water molecule in the liquid state?
 a. Liquid
 b. Solid
 c. Gas
 d. Liquid
6. What is the liquid state of water?
 a. 100°C
 b. 100°C
 c. 100°C
 d. 100°C
7. A molecule's chemical formula is H_2O , at the same time it is 100% hydrogen bonding network.
 a. No
 b. Yes
 c. No
 d. Yes
8. Pure water at T_m (100°C) is 100% hydrogen bonding network (hydrogen bonding network is 100% at 100°C).
 a. No
 b. Yes
 c. No
 d. Yes
9. At standard T_m (100°C) water has a hydrogen bonding network.
 a. Yes
 b. No
 c. Yes
 d. No
10. The hydrogen bond donor in the liquid state:
 a. Don't do anything
 b. Give electrons
 c. Give electronegativity
 d. Give electronegative
11. The hydrogen bond donor in the liquid state:
 a. Don't do anything
 b. Give electrons
 c. Give electronegativity
 d. Give electronegative

Multiple Choice Stoichiometry Test

Mary Millhollon, Richard H. Langley



Multiple Choice Stoichiometry Test:

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