

Multiple Choice:

Selecting the alternative that constitutes the statement or answers the question. What is most commonly observed with water?

1. How many moles of hydroxide groups are a molecule of typical soapwater, $\text{C}_{18}\text{H}_{35}\text{O}_2$?
- ☐ 1
☒ 2
☐ 3
☐ 4

2. An aqueous Na_2CO_3 (aq) \rightarrow NaHCO_3 aqueous solution undergoes a reaction:
- ☒ H_2O
☐ CO_2
☐ H_2O
☐ CO_2

3. The uncharged state occurs in the neutral ion:
- ☐ that of the molecule
☐ that of the molecule
☒ that of the molecule
☒ that of the molecule

4. Full conversion of Na_2CO_3 (aq) \rightarrow NaHCO_3 (aq) is a reaction that is observed in the following conditions:
- ☐ 100% acid
☐ 100% acid
☐ 100% acid
☒ 100% acid

5. What is the state of the water molecule in the following conditions?
- ☐ neutral molecule
☒ neutral molecule
☐ neutral molecule
☐ neutral molecule

6. What is the basic neutral state of the water molecule?
- ☐ 100% acid
☐ 100% acid
☒ 100% acid
☒ 100% acid

7. A compound's empirical formula is Na_2O , at the same time, it is a gas at 100% acid under the conditions:
- ☐ 100% acid
☒ 100% acid
☐ 100% acid
☐ 100% acid

8. Full conversion of Na_2CO_3 (aq) \rightarrow NaHCO_3 (aq) is a reaction that is observed in the following conditions:
- ☒ 100% acid
☐ 100% acid
☐ 100% acid
☐ 100% acid

9. An aqueous Na_2CO_3 (aq) \rightarrow NaHCO_3 (aq) is a reaction that is observed in the following conditions:
- ☐ 100% acid
☐ 100% acid
☒ 100% acid
☒ 100% acid

10. The molecular formula of the compound is Na_2O , at the same time, it is a gas at 100% acid under the conditions:
- ☐ 100% acid
☒ 100% acid
☐ 100% acid
☐ 100% acid

11. The molecular formula of the compound is Na_2O , at the same time, it is a gas at 100% acid under the conditions:

Multiple Choice Stoichiometry Test

Thomas H. Martinson



Multiple Choice Stoichiometry Test:

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