

14.3

HYDROGEN IONS AND ACIDITY

Section Review

Objectives

1. Calculate concentrations based on the pH of acids and bases and vice versa.
2. Calculate pH and concentrations based on K_a and K_b and vice versa.
3. Calculate the pH of weak acids and bases.

Vocabulary

- | | | |
|------------------------|-----------------------------|-------------------|
| 1. autoionization | 2. conjugate acid-base pair | 3. weak acid/base |
| 4. ionization | 5. K_a | 6. K_b |
| 7. ionization constant | 8. pH | |

Key Equation

- | | |
|-----------------------|-----------------------|
| 1. $[H^+] = 10^{-pH}$ | 2. $pH = -\log [H^+]$ |
| 3. $pH = -\log [H^+]$ | 4. $[H^+] = 10^{-pH}$ |

Part A Completion

Write the equation and/or the numerical value(s) of the constant(s) that are indicated in the spaces provided. Be sure to include proper units when appropriate.

1. Water autoionizes. K_w is the ion-product constant for water. Write the equation and K_w at 25°C.
2. The autoionization of water is endothermic. Write the equation and K_w at 50°C.
3. The autoionization of water is endothermic. Write the equation and K_w at 0°C.
4. The autoionization of water is endothermic. Write the equation and K_w at 100°C.
5. The autoionization of water is endothermic. Write the equation and K_w at 150°C.
6. The autoionization of water is endothermic. Write the equation and K_w at 200°C.
7. The autoionization of water is endothermic. Write the equation and K_w at 250°C.
8. The autoionization of water is endothermic. Write the equation and K_w at 300°C.
9. The autoionization of water is endothermic. Write the equation and K_w at 350°C.
10. The autoionization of water is endothermic. Write the equation and K_w at 400°C.
11. The autoionization of water is endothermic. Write the equation and K_w at 450°C.
12. The autoionization of water is endothermic. Write the equation and K_w at 500°C.
13. The autoionization of water is endothermic. Write the equation and K_w at 550°C.
14. The autoionization of water is endothermic. Write the equation and K_w at 600°C.
15. The autoionization of water is endothermic. Write the equation and K_w at 650°C.
16. The autoionization of water is endothermic. Write the equation and K_w at 700°C.
17. The autoionization of water is endothermic. Write the equation and K_w at 750°C.
18. The autoionization of water is endothermic. Write the equation and K_w at 800°C.
19. The autoionization of water is endothermic. Write the equation and K_w at 850°C.
20. The autoionization of water is endothermic. Write the equation and K_w at 900°C.
21. The autoionization of water is endothermic. Write the equation and K_w at 950°C.
22. The autoionization of water is endothermic. Write the equation and K_w at 1000°C.

1. $K_w = 1.0 \times 10^{-14}$
2. $K_w = 1.0 \times 10^{-14}$
3. $K_w = 1.0 \times 10^{-14}$
4. $K_w = 1.0 \times 10^{-14}$
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21. $K_w = 1.0 \times 10^{-14}$
22. $K_w = 1.0 \times 10^{-14}$

19 2 Hydrogen Ions And Acidity Work Sheet Answers

Puja Mehta



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