

Multiple Choice:

Selecting the alternative that constitutes the statement or answers the question. What is most commonly observed with water?

1. How many moles of hydrogen gas are in a cylinder at typical conditions, 1000 L?
 a. 1
 b. 10
 c. 100
 d. 1000

2. At standard T_p , 1 mol of H_2O vapor has higher enthalpy than liquid water.
 a. True
 b. False
 c. Not sure
 d. None

3. The enthalpy of fusion is the amount of:
 a. Heat of the system
 b. Heat of the surroundings
 c. Heat of the universe
 d. Heat of the system and surroundings

4. First law statement: $\Delta U = q + w$. If $q = 0$, $\Delta U = w$. In a closed system, if $q = 0$, $\Delta U = w$. In a closed system, if $q = 0$, $\Delta U = w$. In a closed system, if $q = 0$, $\Delta U = w$.
 a. Not true
 b. True
 c. Not sure
 d. None

5. What is the study of the relationship between mechanical work and heat?
 a. Thermodynamics
 b. Kinetics
 c. Equilibrium
 d. None

6. What is the heat of fusion of water?
 a. 1.0 kJ/mol
 b. 1.0 kJ/g
 c. 1.0 kJ/kg
 d. 1.0 kJ/L

7. A substance's molar heat of fusion is 10 kJ/mol. At this point, the substance is at its melting point. The substance is at its melting point.
 a. True
 b. False
 c. Not sure
 d. None

8. First law statement: $\Delta U = q + w$. If $q = 0$, $\Delta U = w$. In a closed system, if $q = 0$, $\Delta U = w$. In a closed system, if $q = 0$, $\Delta U = w$.
 a. True
 b. False
 c. Not sure
 d. None

9. At standard T_p , 1 mol of H_2O vapor has higher enthalpy than liquid water.
 a. True
 b. False
 c. Not sure
 d. None

10. The enthalpy of fusion of water is 10 kJ/mol. At this point, the substance is at its melting point.
 a. True
 b. False
 c. Not sure
 d. None

11. The enthalpy of fusion of water is 10 kJ/mol. At this point, the substance is at its melting point.
 a. True
 b. False
 c. Not sure
 d. None

Multiple Choice Stoichiometry Test

Mary Millhollon, Richard H. Langley



Multiple Choice Stoichiometry Test:

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