

Multiple Choice

Select the alternative that best completes the statement or answers the question. What is the most common primary gas used?

1. How many moles of hydrogen gas are in a cylinder at 100 atm pressure, 20°C?
- ☐ 1
☒ 2
☐ 3
☐ 4

2. At constant $T_1 = 100^\circ\text{C}$, what is the molar volume of hydrogen gas?
- ☒ 20
☐ 10
☐ 5
☐ 1

3. The molar volume of hydrogen is the same as:
- ☐ that of any gas
☐ that of any liquid
☒ that of any solid
☒ that of any gas

4. For hydrogen, $T_1 = 100^\circ\text{C}$, $T_2 = 20^\circ\text{C}$, $P_1 = 100$ atm, $P_2 = 1$ atm, what is the change in internal energy?
- ☐ 0 J/mol
☐ 10 J/mol
☐ 20 J/mol
☒ 30 J/mol

5. What is the state of the hydrogen gas?
- ☐ liquid
☐ solid
☒ gas
☒ liquid

6. What is the final molar volume?
- ☐ 10 J/mol
☐ 20 J/mol
☐ 30 J/mol
☒ 40 J/mol

7. A cylinder of hydrogen gas is at 100 atm, at the same time, it is at 20°C. What is the final molar volume?
- ☒ 20
☐ 10
☐ 5
☐ 1

8. For hydrogen, $T_1 = 100^\circ\text{C}$, $T_2 = 20^\circ\text{C}$, $P_1 = 100$ atm, $P_2 = 1$ atm, what is the change in internal energy?
- ☒ 20 J/mol
☐ 10 J/mol
☐ 5 J/mol
☐ 1 J/mol

9. At constant $T_1 = 100^\circ\text{C}$, $T_2 = 20^\circ\text{C}$, $P_1 = 100$ atm, $P_2 = 1$ atm, what is the change in internal energy?
- ☐ 10
☐ 20
☒ 30
☐ 40

10. What is the final molar volume?
- ☐ 10 J/mol
☒ 20 J/mol
☐ 30 J/mol
☐ 40 J/mol

11. The molar volume of hydrogen is the same as:
- ☐ that of any gas
☐ that of any liquid
☐ that of any solid
☐ that of any gas

Multiple Choice Stoichiometry Test

Mary Millhollon, Richard H. Langley



Multiple Choice Stoichiometry Test:

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