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# PERSONAL PROTECTIVE EQUIPMENT (PPE)

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## STUDY GUIDE

OSHA 10-HOUR (CONSTRUCTION)

MODULE 08/11



# Osha Personal Protective Equipment Self Study Guide

**Office of the Federal Register**

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web donor whereas a base generically represented by b is a proton acceptor when acids and bases combine they swap a proton ha b a hb strong acids e g sulfuric acid have a very pronounced tendency to transfer a proton to a base whereas weak acids

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web understand that parts of an acid base titration be able to determine the  $K_a$  or  $K_b$  from ph data associated with the titration of a weak acid or base be able to determine the molar mass of a solid monoprotic acid from titration data be able to calculate  $K_{a1}$  and  $K_{a2}$  for a polyprotic acid by the end of this lab students should be able to

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web 1 approximately 3 g of hydrated oxalic acid h<sub>2</sub>c<sub>2</sub>o<sub>4</sub>2h<sub>2</sub>o was weighted to the nearest 0 g in a 50 ml beaker 2 about 30 ml of distilled water was added to the beaker to dissolve the oxalic acid 3 the solution was transferred into 250 ml volumetric flask the beaker was rinsed and its contents were poured into the flask

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