

14.3

HYDROGEN IONS AND ACIDITY

Section Review

Objectives

1. Calculate concentrations based on the pH of acids and bases and vice versa.
2. Calculate pH values and concentrations based on the pH and K_a or K_b and vice versa.
3. Calculate the pH of weak acids and bases.

Vocabulary

- | | | |
|-------------------|-----------------------------|---------------------|
| 1. autoionization | 2. conjugate acid-base pair | 3. weak acid/base |
| 4. autoionization | 5. acid rain | 6. relative acidity |
| 7. autoionization | 8. acid rain | 9. pH |

Key Equation

- | | |
|-------------------------------------|-------------------------------------|
| 1. $\text{pH} = -\log [\text{H}^+]$ | 2. $\text{pH} = -\log [\text{H}^+]$ |
| 3. $\text{pH} = -\log [\text{H}^+]$ | 4. $\text{pH} = -\log [\text{H}^+]$ |

Part A Completion

Write the equation and/or the numerical value(s) for the reaction(s) described. Show the calculation for pH or percent dissociation when required. Write your answer(s) in the space provided.

1. What substance is H^+ in water? Hydrogen ion.
2. What substance is OH^- in water? Hydroxide ion.
3. What substance is H^+ in acid rain? Hydrogen ion.
4. What substance is OH^- in acid rain? Hydroxide ion.
5. What substance is H^+ in acid rain? Hydrogen ion.
6. What substance is OH^- in acid rain? Hydroxide ion.
7. What substance is H^+ in acid rain? Hydrogen ion.
8. What substance is OH^- in acid rain? Hydroxide ion.
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14. What substance is OH^- in acid rain? Hydroxide ion.
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16. What substance is OH^- in acid rain? Hydroxide ion.
17. What substance is H^+ in acid rain? Hydrogen ion.
18. What substance is OH^- in acid rain? Hydroxide ion.
19. What substance is H^+ in acid rain? Hydrogen ion.
20. What substance is OH^- in acid rain? Hydroxide ion.

1. H^+
2. OH^-
3. H^+
4. OH^-
5. H^+
6. OH^-
7. H^+
8. OH^-
9. H^+
10. OH^-
11. H^+
12. OH^-
13. H^+
14. OH^-
15. H^+
16. OH^-
17. H^+
18. OH^-
19. H^+
20. OH^-

19 2 Hydrogen Ions And Acidity Worksheet Answers

Yijin Wang



19 2 Hydrogen Ions And Acidity Worksheet Answers:

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