

## 19.2 Hydrogen Ions and Acidity

- The reaction in which water molecules produce ions is called the self-ionization of water
- $\text{H}_2\text{O(l)} \rightleftharpoons \text{H}^+(\text{aq}) + \text{OH}^-(\text{aq})$
- $2\text{H}_2\text{O(l)} \rightleftharpoons \text{H}^3\text{O}^+(\text{aq}) + \text{OH}^-(\text{aq})$



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**Terry C. Jones**

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In a world inundated with displays and the cacophony of immediate conversation, the profound energy and emotional resonance of verbal artistry frequently disappear into obscurity, eclipsed by the continuous assault of noise and distractions. Yet, set within the lyrical pages of **19 2 Hydrogen Ions And Acidity**, a captivating work of literary elegance that impels with raw thoughts, lies an memorable trip waiting to be embarked upon. Composed with a virtuoso wordsmith, that exciting opus courses visitors on a psychological odyssey, softly revealing the latent possible and profound affect embedded within the complex web of language. Within the heart-wrenching expanse with this evocative examination, we shall embark upon an introspective exploration of the book is main themes, dissect their fascinating writing design, and immerse ourselves in the indelible effect it leaves upon the depths of readers souls.

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