

Multiple Choice:

Selecting the alternative that constitutes the statement or answers the question. What is most commonly observed with water?

1. How many moles of hydrogen gas are in a cylinder at typical conditions, 1000 L?
 a. 1
 b. 10
 c. 100
 d. 1000

2. At temperature $T_1 = 100^\circ\text{C}$, water has a vapor pressure of 1000 mmHg.
 a. 100
 b. 1000
 c. 10000
 d. 100000

3. The enthalpy of fusion is the amount of:
 a. heat of the system
 b. heat of the universe
 c. heat of the surroundings
 d. heat of the system

4. For the reaction $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$, the heat of reaction is 285 kJ/mol. The heat of reaction for the reaction $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$ is 285 kJ/mol.
 a. 285 kJ/mol
 b. 570 kJ/mol
 c. 1135 kJ/mol
 d. 11350 kJ/mol

5. What is the state of the water in the atmosphere?
 a. liquid
 b. solid
 c. gas
 d. plasma

6. What is the heat of fusion of water?
 a. 333 J/mol
 b. 333 kJ/mol
 c. 3330 J/mol
 d. 3330 kJ/mol

7. A substance's specific heat is 100 J/kg. At the same time, it is 100 J/kg. The substance is water.
 a. 100 J/kg
 b. 1000 J/kg
 c. 10000 J/kg
 d. 100000 J/kg

8. For the reaction $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$, the heat of reaction is 285 kJ/mol. The heat of reaction for the reaction $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$ is 285 kJ/mol.
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9. At the pressure $P_1 = 1000 \text{ mmHg}$, water has a vapor pressure of 1000 mmHg.
 a. 100
 b. 1000
 c. 10000
 d. 100000

10. The enthalpy of fusion is the amount of:
 a. heat of the system
 b. heat of the universe
 c. heat of the surroundings
 d. heat of the system

11. The enthalpy of fusion is the amount of:
 a. heat of the system
 b. heat of the universe
 c. heat of the surroundings
 d. heat of the system

Multiple Choice Stoichiometry Test

Richard H. Langley

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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