

Multiple Choice

Selecting the alternative that constitutes the statement or answers the question. What is most commonly observed with water?

1. How many moles of hydrogen gas are in a cylinder at typical conditions of use?
- ☐ A. 1
☒ B. 2
☐ C. 3
☐ D. 4

2. For the reaction $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$, what factor determines whether a reaction occurs?
- ☒ A. ΔG
☐ B. ΔH
☐ C. ΔS
☐ D. ΔT

3. The enthalpy of fusion is the amount of:
- ☐ A. heat at the melting point
☐ B. heat of the substance
☒ C. heat of fusion
☐ D. heat of vaporization

4. For the reaction $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$, the free energy change is -237 kJ/mol . How many moles of H_2 are produced per mole of O_2 consumed at this condition?
- ☐ A. 1/2 mole
☐ B. 1 mole
☒ C. 2 moles
☐ D. 4 moles

5. What is the study of the interrelationships among environmental problems in a natural system?
- ☐ A. systems engineering
☒ B. systems engineering
☐ C. systems engineering
☐ D. systems engineering

6. What is the basic natural distribution?
- ☐ A. 1/2 mole
☐ B. 1 mole
☒ C. 2 moles
☐ D. 4 moles

7. A compound's empirical formula is CH_2 , and the molar mass is 42 g/mol. What is the molecular formula?
- ☒ A. C_2H_4
☐ B. C_3H_6
☐ C. C_4H_8
☐ D. C_5H_{10}

8. For the reaction $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$, the free energy change is -237 kJ/mol . How many moles of H_2 are produced per mole of O_2 consumed at this condition?
- ☒ A. 1/2 mole
☐ B. 1 mole
☐ C. 2 moles
☐ D. 4 moles

9. For the reaction $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$, the free energy change is -237 kJ/mol . How many moles of H_2 are produced per mole of O_2 consumed at this condition?
- ☐ A. 1/2 mole
☐ B. 1 mole
☒ C. 2 moles
☐ D. 4 moles

10. The molecular formula of a compound is C_2H_4 . What is the empirical formula?
- ☐ A. C_2H_4
☒ B. CH_2
☐ C. C_4H_8
☐ D. C_6H_{12}

11. The molecular formula of a compound is C_2H_4 . What is the empirical formula?

Multiple Choice Stoichiometry Test

Angela Woodward Spangenberg

A red circular graphic with a gradient, appearing as a partial circle or a thick arc, located to the right of the author's name.

Multiple Choice Stoichiometry Test:

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