

The curve in Figure 16.11 of the text shows how the vibrational heat capacity depends on temperature. Note that even when the temperature is only slightly above the vibrational temperature, the heat capacity is close to its equipartition value.

The total heat capacity of a molecular substance is the sum of each contribution (Figure 16.12 of the text). When equipartition is valid (when the temperature is well above the characteristic temperature of the mode  $T \gg \theta_m$ ) we can estimate the heat capacity by counting the numbers of modes that are active. In gases, all three translational modes are always active and contribute  $\frac{3}{2}R$  to the molar heat capacity. If we denote the number of active rotational modes by  $v_f^*$  (so for most molecules at normal temperatures  $v_f^* = 2$  for linear molecules, and 3 for non-linear molecules), then

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**RJ Shavelson**

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web the t6 sirkeci kazlıçeşme or sirkeci kazlıçeşme rail line turkish sirkeci kazlıçeşme raylı sistemi is a tram line rail line or light rail line that will operate as a shuttle between the sirkeci terminal and kazlıçeşme station the t6 line will use 8 km of track previously used by the istanbul sirkeci pythio railway and istanbul suburban closed