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web 1 approximately 3 g of hydrated oxalic acid h<sub>2</sub>c<sub>2</sub>o<sub>4</sub> 2h<sub>2</sub>o was weighted to the nearest 0 g in a 50 ml beaker 2 about 30 ml of distilled water was added to the beaker to dissolve the oxalic acid 3 the solution was transferred into 250 ml volumetric flask the beaker was rinsed and its contents were poured into the flask

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