

Section 1: Introduction

Because of the high concentration of the product, the reaction mixture was extremely viscous. The reaction mixture with the addition of the starting ether dissolved in a solution of LiClO_4 and $\text{CH}_3\text{CO}_2\text{Et}$ was then deoxygenated. After a 10-min. period, the reaction mixture was heated to 100°. The dry ice was then added and the ether solution was collected. The product was purified by column chromatography on alumina. The product was a colorless, viscous, ether-soluble oil. The infrared spectrum was identical with that of the product obtained from the reaction of $\text{CH}_3\text{CO}_2\text{Et}$ and $\text{CH}_3\text{CO}_2\text{Et}$ in the presence of LiClO_4 and $\text{CH}_3\text{CO}_2\text{Et}$.

11. *What is the name of the author of the book you are reading?*

11. *What is the name of the author of the book?*

For more information, contact the National Institute of Child Health and Human Development (NICHD) at 301-435-0911 or visit the NICHD website at www.nichd.nih.gov.

在這一個月的時間裏，我會盡力地去完成這項工作，並希望得到大家的指導和建議。

• **Conclusions** and **recommendations** for the **future** of the **industry** and **the** **sector**

For a more detailed analysis see [Brennan et al. \(2009\)](#).

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