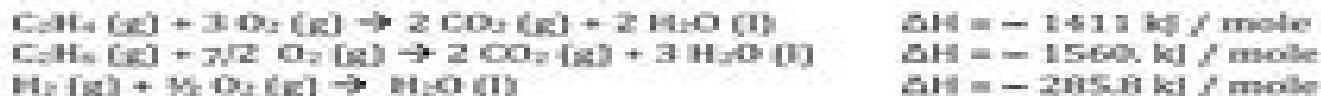


CHEM 12: Hess's Law Worksheet

1. Calculate the ΔH for the reaction: $\text{C}_2\text{H}_6(\text{g}) + \text{H}_2(\text{g}) \rightarrow \text{C}_2\text{H}_4(\text{g})$, from the following data:



2. Calculate the ΔH for the reaction:



3. Find the ΔH_f° (heat of formation) for acetic acid, $\text{HC}_2\text{H}_3\text{O}_2$, using the following thermochemical data:



Pure Factor $\text{C}(\text{s})$ is in the form of graphite.

Acetic acid, $\text{HC}_2\text{H}_3\text{O}_2$, actually has the structure: CH_3COOH



4. Calculate the ΔH for the reaction: $\text{CH}_4(\text{g}) + \text{NH}_3(\text{g}) \rightarrow \text{HCN}(\text{g}) + 3\text{H}_2(\text{g})$, from the following data:



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Alton J. Banks, Paul Treichel

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Thermodynamics Ricardo Morales-Rodriguez,2012-10-03 This book presents the selection of various high level contributions involving thermodynamics The book goes from the fundamentals up to several applications in different scientific fields The content of the book has been classified in six sections Classical Thermodynamics Statistical Thermodynamics Property Prediction in Thermodynamics Material and Products Non Equilibrium and Thermodynamics in Diverse Areas The classification of the book aims to provide to the reader the facility of finding the desired topic included in the book It is expected that this collection of chapters will contribute to the state of the art in the thermodynamics area

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Alton J. Banks,Paul Treichel,1999 *Solutions to Exercises* Roxy Wilson,2002-05 Full solutions to all end of chapter exercises in the text are provided With an instructor s permission this manual may be made available to students

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