

Multiple Choice:

Selecting the alternative that constitutes the statement or answers the question. What is most commonly observed with water?

1. How many moles of hydrogen gas are in a cylinder at typical conditions, 1000 L?  
 a. 1  
 b. 10  
 c. 100  
 d. 1000

2. At temperature  $T_1 = 100^\circ\text{C}$ , water has a vapor pressure of 1000 mmHg.  
 a. 100  
 b. 1000  
 c. 10000  
 d. 100000

3. The condensed phase is in the lowest energy state.  
 a. Most of the time  
 b. Only at low temperatures  
 c. Only at high temperatures  
 d. Only at intermediate temperatures

4. For temperature  $T_1 = 100^\circ\text{C}$ ,  $\Delta H_{\text{vap}} = 40.7 \text{ kJ/mol}$ , how many moles of water vapor are produced when 1000 J of heat is added?  
 a. 100 moles  
 b. 1000 moles  
 c. 10000 moles  
 d. 100000 moles

5. Water is the most abundant liquid on Earth.  
 a. Because of its high boiling point  
 b. Because of its high melting point  
 c. Because of its high density  
 d. Because of its high specific heat

6. Water is the most abundant liquid on Earth.  
 a. Because of its high boiling point  
 b. Because of its high melting point  
 c. Because of its high density  
 d. Because of its high specific heat

7. A substance's condensed phase is the most stable at low temperatures.  
 a. True  
 b. False  
 c. Only at high temperatures  
 d. Only at low temperatures

8. For temperature  $T_1 = 100^\circ\text{C}$ ,  $\Delta H_{\text{vap}} = 40.7 \text{ kJ/mol}$ , how many moles of water vapor are produced when 1000 J of heat is added?  
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10. The condensed phase is the most stable at low temperatures.  
 a. True  
 b. False  
 c. Only at high temperatures  
 d. Only at low temperatures

11. The condensed phase is the most stable at low temperatures.  
 a. True  
 b. False  
 c. Only at high temperatures  
 d. Only at low temperatures

# Multiple Choice Stoichiometry Test

**Rosemary Luckin, Kenneth R.  
Koedinger, Jim E. Greer**



## Multiple Choice Stoichiometry Test:

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