
Identify the administrative controls that facilitate or impede the process. (Fig. 8) (not necessary, depending on case)

9. If the energy content of carbohydrates present in a molecule is 4 kcal/g, carbohydrates with $\text{C}_{10}\text{H}_{18}\text{O}_5$ will have an energy content of _____ kcal/g.
 a. 1
 b. 2
 c. 4
 d. 8
10. For carbohydrates $\text{C}_n(\text{H}_2\text{O})_n$, $n = 1000$, what factor explains why they are considered macromolecules?
 a. 100
 b. 1000
 c. 10000
 d. 100000
11. The endogenous effect observed in the bacterial cell:
 a. shut off gene expression
 b. gene expression
 c. shut off transcription
 d. gene expression
12. For carbohydrates $\text{C}_n(\text{H}_2\text{O})_n$, $n = 1000$, what factor explains why they are considered macromolecules?
 a. 100
 b. 1000
 c. 10000
 d. 100000
13. What is the study of the order in biological systems?
 a. chemical composition
 b. biological order
 c. biological structure
 d. biological organization
14. What is the basic building block of DNA?
 a. 5' deoxyribose
 b. 3' deoxyribose
 c. 5' ribose
 d. 3' ribose
15. A compound's empirical formula is C_2H_4 . At the molecular level, it is 50 g/mol. What is the molecular formula?
 a. C_2H_4
 b. C_5H_{10}
 c. $\text{C}_{10}\text{H}_{20}$
 d. $\text{C}_{20}\text{H}_{40}$
16. For carbohydrates $\text{C}_n(\text{H}_2\text{O})_n$, $n = 1000$, what factor explains why they are considered macromolecules?
 a. 100
 b. 1000
 c. 10000
 d. 100000
17. For carbohydrates $\text{C}_n(\text{H}_2\text{O})_n$, $n = 1000$, what factor explains why they are considered macromolecules?
 a. 100
 b. 1000
 c. 10000
 d. 100000
18. The molecular formula for glucose is $\text{C}_6\text{H}_{12}\text{O}_6$. What is the empirical formula?
 a. $\text{C}_6\text{H}_{12}\text{O}_6$
 b. $\text{C}_3\text{H}_6\text{O}_3$
 c. $\text{C}_2\text{H}_4\text{O}_2$
 d. CH_2O
19. The molecular formula for glucose is $\text{C}_6\text{H}_{12}\text{O}_6$. What is the empirical formula?
 a. $\text{C}_6\text{H}_{12}\text{O}_6$
 b. $\text{C}_3\text{H}_6\text{O}_3$
 c. $\text{C}_2\text{H}_4\text{O}_2$
 d. CH_2O

Multiple Choice Stoichiometry Test

Michael Seilmaier

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the name bar.

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- Chapter 3: Multiple Choice Stoichiometry Test in Everyday Life
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