



**BUSINESS STUDIES**  
**GRADE 10**  
**PAPER 2**  
**TERM 2**  
**CHAPTER 10 (PART 2)**  
**FORMS OF OWNERSHIP**  
**REVISED NOTES**  
**2024**

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web in the context of a chemical reaction the term neutralization is used for a reaction between an acid and a base or alkali historically this reaction was represented as acid base alkali salt water for example  $\text{hcl naoh nacl h}_2\text{o}$

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web reacting two solutions eg acid and alkali place the polystyrene cup inside the glass beaker to make it more stable measure an appropriate volume of each liquid eg 25 cm<sup>3</sup> place one of the

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web for reactions involving acetic acid or ammonia the measured enthalpy change of neutralization is a few kJ less exothermic than with strong acids and bases for example one source which gives the enthalpy change of neutralization of

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web sep 20 2014 i will illustrate this with some examples when an acid is neutralised by an alkali the reaction is exothermic eg 1 h cl aq n aoh aq n aql aq h 2o l for which  $\Delta h$  57kj mol 1 eg 2 h n o3 aq koh aq kn o3 aq h 2o l for which  $\Delta h$  57kj mol 1

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