

### Chapter 1

1. What did Annie think she saw in the woods?

.....

2. Why did Jack say that they needed to go home?

.....

3. What was the tree house full of?

.....

### Chapter 2

1. There were two kinds of books in the tree house. Name and describe the two kinds. ....

.....

2. What kind of dinosaur did Jack find in the book?

.....

3. What was the monster flying outside the tree house? .....

.....

### Chapter 3

1. What did Jack see in the distance of the new land?

.....

# **Magic Tree House Comprehension Questions**

**Marcia Amidon Lusted**



## **Magic Tree House Comprehension Questions:**

Reading with the Magic Treehouse Gr. 1-3 , **The Viking Princess (Novel Study) Gr. 5-8** Nat Reed,2013-01-13 The Viking Princess is about a twelve year old girl named Edith Valgardsdottir who is expected to tend the family's sheep and clear rocks from her father's fields But Edith dreams of exploring glorious new worlds and becoming a leader of her people Trapped aboard a Viking ship Edith suddenly has the unwelcomed opportunity of being among the first of her village to explore the far off coast of Markland Labrador That is if she can survive her first encounters with the crew of the Signe Capture first hand the flavor and spirit of life with the Norsemen This Novel Study provides a teacher and student section with a variety of activities chapter questions crossword word search and answer key to create a well rounded lesson plan

**Samurai Science** Marcia Amidon Lusted,2017-08-10 Describes the science behind the armor weapons training and tactics used by Samurai warriors in combat Teaching Literacy in Third Grade Janice F. Almasi,Keli Garas-York,Leigh-Ann Hildreth,2013-10-21 The third grade classroom is a beehive of activity in which young readers transition between emergent and more advanced levels of literacy This expertly written guide brings to life the rewards and challenges of teaching third graders and helps teachers differentiate instruction for diverse learners Vividly portraying a week in a highly motivating classroom the authors present easy to use ideas and activities for building fluency vocabulary comprehension writing and more Illustrations reproducibles grade specific resources and planning tips will make this handy book a boon to third grade teachers every day of the year

**Ninja Science** Marcia Amidon Lusted,2017-08-10 Ninja warriors were masters of stealth and disguise They could quickly defeat their enemies without raising the alarm They may not have realized it but they owed their success to science Engaging text fascinating photos and useful illustrations will help young readers understand how science helped ninjas achieve success in their secret missions Difficult Texts and the Students who Choose Them Juliet L. Halladay,2008 Quantum Pad Smapler - Included Book ,2002 A Guide for Using Harry Potter and the Sorcerer's Stone/Other Harry Potter Books in the Classroom ,2002-06 Teaching literature unit based on the popular children's story Harry Potter and the sorcerer's stone Smart Guide to Third Grade ,2002 Get ready to learn what you need to know in 3rd grade Interactive games make learning easy and fun Page 4 of cover of accompanying book

**Reading Problems** Margaret Ann Richek,Lynne K. List,Janet W. Lerner,1983 The Knight at Dawn ,2002 Eight year old Jack and his younger sister Annie use the magic treehouse to travel back to the Middle Ages where they explore a castle and are helped by a mysterious knight Democracy & Education ,1994 **Transforming Lives** Michael Gerald Luther,2007 **Making Movies** ,2002 Go behind the scenes in the movie industry meet celebrities mix your own soundtrack learn about the history of film and more **School Library Journal** ,2010 Proceedings ,2003 **--and with a Light Touch** Carol Avery,2002

It's my bible for teaching young children and It reads like a novel said veteran and preservice teachers alike *Reading Mastery Classic Grades Pre-K-2, Series Guide* McGraw Hill,2002-07-11 For more than thirty five years Reading Mastery

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**The Bookmark Reading Program: Preprimer I: Sun up** Margaret Early,1968      *Competency in Teaching Reading* Carl J. Wallen,1981

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the objectives of the boiling point determination lab were to determine the boiling point of a mystery substance provided to each student boiling point is the point at which a liquid changes physical states and becomes a vapor or gas boiling point is important to understand because it can be used to help distinguish between different substances

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the boiling point of an organic liquid is the temperature at which its vapour pressure equals the atmospheric pressure over the liquid or it is the temperature at which the vapour and liquid phases are in equilibrium at a given pressure the boiling point is considered as a criterion of purity of a compound and is useful for identification of

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given that the boiling points of ethanol and methanol 78 c and 64 c respectively are far off in a significant way this seems like a valid conclusion conclusion thermometer calibrations are necessary in order to collect reliable information as precise calculations are imperative to gathering correct results

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melting point 1 a push the open end of a capillary tube into the powdered acetamide b move the powder to the closed end of the capillary tube by tapping it on the table repeat until the powdered acetamide occupies 1 2 mm of the capillary tube end c

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discussion and conclusion the technique recrystallization was used because it is a simple method to purify a solid compound the percentage yield for benzoic acid and acetanilide acid being high indicated the initial sample had a low concentration of impurities with the melting point being in a specific

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determination of boiling points introduction the boiling point of a compound is the temperature at which it changes from a

liquid to a gas this is a physical property often used to identify substances or to check the purity of the compound it

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clean up solid and liquid spills immediately true g conclusions after conducting the boiling point determination lab it can be concluded that the unknown substance 193 is isopropyl alcohol on account of its observed boiling point of 88 c

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conclusion after completing this experiment the melting point of three different unknown compounds was obtained using the mel temp apparatus compound a had a melting point range of 114 6 115 5 c thus revealing it to be acetanilide compound b had a melting point range of 122 6 124 5 c thus

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nov 27 2019 essay sample the purpose of this experiment is to determine the boiling points of various organic compounds and to use these to identify unknowns equipment free essays

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melting point can be used as the identity and purity of an organic solid however it is not the same as the boiling point the boiling point is a characteristic unique of pure liquids the boiling point is the temperature at which the vapor pressure of a liquid is equal to the atmospheric pressure

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exploring boiling points abstract in this experiment we are given a set of straight chain alkanes and a set of compounds to compare by following the trendline of the straight chain alkanes it could be concluded that as the molar mass of the straight chain alkanes increases their boiling points increase as well

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50 c so in order to convert an experimental measurement taken in calgary which is at higher altitude than sea level and so is at a lower pressure to that reported for sea level higher pressure one needs to add a correction factor since the boiling point at sea level is higher than that at higher altitudes

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introduction in chemistry one may encounter an unknown substance that need to be determined by physical properties such as boiling point melting point density solubility etc in this experiment i am to figure out the boiling point of a mystery substance in order to identify it chemical responsibility this experiment requires the handling of a bunsen burner boiling

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there are a variety of methods by which a sample s boiling point can be determined including distillation reflux and by using a thiele tube the most straightforward method uses a thiele tube and has the advantage of using less than 0 5ml 0 5 ml of material

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