

Multiple Choice

Selecting the alternative that constitutes the statement or answers the question. What is most commonly observed with water?

1. How many moles of hydroxide groups are a molecule of typical silicate tetrahedron?
 a. 1
 b. 2
 c. 3
 d. 4

2. For the reaction $\text{Si}_2\text{O}_2 + 2\text{H}_2\text{O} \rightarrow 2\text{SiOH}_2$, what factor determines whether a reaction?
 a. ΔG
 b. ΔH
 c. ΔS
 d. ΔG°

3. The unhydrated silicate anion in the fused sil.
 a. SiO_4^{4-}
 b. SiO_3^{2-}
 c. SiO_2
 d. SiO_4^{2-}

4. For the reaction $\text{Si}_2\text{O}_2 + 2\text{H}_2\text{O} \rightarrow 2\text{SiOH}_2$, what factor determines whether a reaction?
 a. ΔG
 b. ΔH
 c. ΔS
 d. ΔG°

5. What is the study of the relationships between mineral products in natural systems?
 a. mineralogy
 b. petrology
 c. mineralogy
 d. mineralogy

6. What is the basic mineral structure?
 a. SiO_4^{4-}
 b. SiO_3^{2-}
 c. SiO_2
 d. SiO_4^{2-}

7. A mineral's empirical formula is Si_2O_2 , at this ratio, what is the ratio of silicon to oxygen?
 a. 1:1
 b. 1:2
 c. 2:1
 d. 2:2

8. For the reaction $\text{Si}_2\text{O}_2 + 2\text{H}_2\text{O} \rightarrow 2\text{SiOH}_2$, what factor determines whether a reaction?
 a. ΔG
 b. ΔH
 c. ΔS
 d. ΔG°

9. For the reaction $\text{Si}_2\text{O}_2 + 2\text{H}_2\text{O} \rightarrow 2\text{SiOH}_2$, what factor determines whether a reaction?
 a. ΔG
 b. ΔH
 c. ΔS
 d. ΔG°

10. What is the basic mineral structure?
 a. SiO_4^{4-}
 b. SiO_3^{2-}
 c. SiO_2
 d. SiO_4^{2-}

11. The unhydrated silicate anion in the fused sil.
 a. SiO_4^{4-}
 b. SiO_3^{2-}
 c. SiO_2
 d. SiO_4^{2-}

Multiple Choice Stoichiometry Test

Rachel S Tattersall



Multiple Choice Stoichiometry Test:

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