

Chapter 3: Water and the Fitness of the Environment

1. Study the water molecules at the right. On the central molecule, label oxygen and hydrogen. ✓

2. What is a polar molecule? Why is water considered polar?

Since oxygen is more electronegative than hydrogen, the electrons of the covalent bonds between them are more strongly attracted to oxygen's than hydrogen's nucleus. This unequal distribution of electrons makes water a polar molecule, meaning that the two ends of the molecule have opposite charges.

3. Add + and - signs to indicate the charged regions of each molecule. Indicate the hydrogen bonds. ✓

4. Explain hydrogen bonding. How many hydrogen bonds can a single water molecule form?

Hydrogen bonds occur when H bonds with F, O, or N and the partially positive H of one molecule is attracted to the partially negative F, O, or N of a nearby molecule. A single water molecule can form four hydrogen bonds.

5. Distinguish between cohesion and adhesion.

Cohesion is the attraction water molecules have towards each other due to hydrogen bonding; adhesion is their attraction to molecules of another substance.

6. What is demonstrated when you see beads of water on a waxed car hood?

The beads have not stretched or broken because of water's high surface tension.

7. Which property explains the ability of a water strider to walk on water?

The point at which the surface of the water interacts with air is an ordered molecular arrangement, causing the water to behave as though coated with an invisible film. This is attributable to water's high surface tension.

8. The calorie is a unit of heat.

A calorie is the amount of heat needed to raise the temperature of 1g of water by 1°C and also the amount of heat released by 1g of water when it cools by 1°C. One calorie is equal to 4.184J.

9. Water has high specific heat. What does this mean? How does water's specific heat compare to alcohol's?

A substance's specific heat is the amount of heat that must be absorbed or lost for 1g of that substance to change its temperature by 1°C. Water's specific heat is 1 cal/g/°C, unusually high compared to other substances. Ethyl alcohol, found in alcoholic beverages, only requires 0.6 calories to raise 1g by 1°C.

10. Explain how hydrogen bonding contributes to water's high specific heat.

Heat must be absorbed in order to break hydrogen bonds, and heat is released when hydrogen bonds form. Specific heat indicates how well a substance resists changing its temperature when it absorbs or releases heat. Due to its molecular structure, water resists temperature change, and absorbs or loses a relatively large quantity of heat for each degree of change.

11. Summarize how water's high specific heat contributes to the moderation of temperature. How is this property important to life?

Large bodies of water can absorb a tremendous amount of heat from the sun while only warming up by a few degrees, while gradually cooling water can warm the air when sunlight is scarce. Thus, in controlling the range of temperature fluctuations through water's high specific heat, bodies of water allow for milder climates on nearby land and stable ocean temperatures, conditions which favor the sustenance of life.

12. Define evaporation. What is heat of vaporization? Explain at least three effects of this property on living organisms.

Evaporation is a substance's transformation from a liquid to a gaseous state. Heat of vaporization is the quantity of heat a liquid must absorb for 1g of it to evaporate. Relative to other liquids, water has a relatively high heat of vaporization due

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web feb 28 2023 badan pusat statistik kabupaten kudos statistics of kudos jl mejobo kompleks perkantoran kudos 59319 telp 62 291 433382 faks 62 21 4251140 mailbox bps3319 bps go id untuk tampilan terbaik anda dapat gunakan berbagai jenis

browser kecuali ie mozilla firefox 3 and safari 3 2 dengan lebar minimum browser beresolusi

perda kab kudus no 9 tahun 2017 jdih bpk ri - Jun 06 2022

web peraturan daerah perda kabupaten kudus nomor 9 tahun 2017 tentang perubahan atas peraturan daerah kabupaten kudus nomor 3 tahun 2015 tentang pedoman penyusunan organisasi dan tata kerja pemerintah desa

perda apbd tahun 2021 ppid kudus kuduskab go id - Apr 04 2022

web perda apbd tahun 2021 data kosong selamat datang di portal resmi ppid pemerintah kabupaten kudus jam pelayanan senin kamis 08 00 15 00 wib jum at 08 00 11 00 wib

bupati kudus provinsi jawa tengah peraturan daerah kabupaten kudus - Mar 15 2023

web 25 peraturan daerah kabupaten kudus nomor 1 tahun 2021 tentang pertanggungjawaban pelaksanaan anggaran pendapatan dan belanja daerah kabupaten kudus tahun anggaran 2020 lembaran daerah kabupaten kudus tahun 2021 nomor 1 dengan persetujuan bersama dewan perwakilan rakyat daerah kabupaten

perda kab kudus no 09 tahun 2022 jdih bpk ri - Jan 13 2023

web peraturan daerah perda bentuk singkat perda tahun 2022 tempat penetapan kudus tanggal penetapan 28 desember 2022 tanggal pengundangan 29 desember 2022 tanggal berlaku 29 desember 2022 sumber ld 2022 no 9 subjek apbd status berlaku bahasa indonesia lokasi pemerintah kabupaten kudus

perda kab kudus no 1 tahun 2022 jdih bpk ri - Oct 22 2023

web peraturan daerah perda kabupaten kudus nomor 1 tahun 2022 tentang rencana tata ruang wilayah kabupaten kudus tahun 2022 2042 t e u indonesia kabupaten kudus nomor 1 bentuk peraturan daerah perda bentuk singkat perda tahun 2022 tempat penetapan kudus tanggal penetapan 31 maret 2022 tanggal pengundangan 31 maret

perda kab kudus no 5 tahun 2022 jdih bpk ri - Sep 21 2023

web peraturan daerah perda kabupaten kudus nomor 5 tahun 2022 tentang perubahan atas peraturan daerah kabupaten kudus nomor 3 tahun 2016 tentang pembentukan dan susunan perangkat daerah kabupaten kudus

jdih kudus - Jul 19 2023

web selamat datang di website bagian hukum setda kabupaten kudus berikut produk hukum terbaru perda nomor 9 tahun 2022 berlaku jl simpang tujuh no 1 kab kudus jawa tengah mail us bagianhukum kuduskab go id 0291 435025 faks

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