

Name: \_\_\_\_\_

## Research Paper Rubric

	4	3	2	1
<b>Content and Focus</b>	<ul style="list-style-type: none"> <li>Exceptionally clear, focused, interesting thesis.</li> <li>Strong, rich supporting details and examples that prove thesis.</li> <li>A meaningful conclusion explaining the importance of the research and how it can be used.</li> </ul>	<ul style="list-style-type: none"> <li>Clear thesis which maintains a consistent focus from beginning to end.</li> <li>Specific supporting details are present.</li> <li>A clear conclusion as to why the research is important.</li> </ul>	<ul style="list-style-type: none"> <li>Contains thesis but with inconsistent focus.</li> <li>Generalized supporting details that prove thesis.</li> <li>Conclusion tends to summarize research.</li> </ul>	<ul style="list-style-type: none"> <li>Thesis statement lacks clarity and focus.</li> <li>Inadequate or missing supporting details.</li> <li>Missing a summarizing conclusion.</li> </ul>
<b>Organization</b>	<ul style="list-style-type: none"> <li>Strong introduction and conclusion.</li> <li>Consistent and coherent logical progression.</li> <li>Uses clear and skillful transitions.</li> </ul>	<ul style="list-style-type: none"> <li>Clear introduction and conclusion.</li> <li>Illustrates some consistency and shows some logical progression.</li> <li>Uses clear transitions.</li> </ul>	<ul style="list-style-type: none"> <li>Introduction and conclusion is present but not clear.</li> <li>Show some attempt of consistency and order. Paper shows attempt of transitions between paragraphs.</li> </ul>	<ul style="list-style-type: none"> <li>Unable to clearly identify introduction and conclusion.</li> <li>Lack of consistency and order.</li> <li>Paper shows little or no attempt of transitions between paragraphs.</li> </ul>
<b>Style</b>	<ul style="list-style-type: none"> <li>Written in formal language (avoids slang completely).</li> <li>Elaborate and colorful language.</li> <li>Consistently strong and varied sentence structure.</li> <li>Direct quotes support student's ideas.</li> <li>Paper written in student's own words.</li> </ul>	<ul style="list-style-type: none"> <li>Majority of paper written in formal language.</li> <li>Language appropriate to topic.</li> <li>Words convey intended message.</li> <li>Direct quotes support student's ideas.</li> <li>Majority of paper written in student's own words.</li> </ul>	<ul style="list-style-type: none"> <li>Some use of formal language recognized; informal language is dominant.</li> <li>Most language is appropriate to topic.</li> <li>Able to get vague idea of message.</li> <li>Some parts of paper written in student's own words.</li> </ul>	<ul style="list-style-type: none"> <li>Paper frequently uses informal language.</li> <li>Language is not appropriate to topic.</li> <li>Message is unclear.</li> <li>Majority of paper is plagiarized.</li> </ul>
<b>Sources/Format</b>	<p>Follows MLA guidelines:</p> <ul style="list-style-type: none"> <li>Uses 5 or more cited sources.</li> <li>Sources meet the guidelines for types of sources.</li> <li>All parenthetical documentation is MLA correct (author's name, pg. #).</li> <li>Works Cited page is MLA correct.</li> <li>All researched info is documented.</li> </ul>	<p>Follows MLA Guidelines with few exceptions:</p> <ul style="list-style-type: none"> <li>3-4 cited sources used.</li> <li>Sources meet the guidelines for types of sources.</li> <li>Few errors noted in parenthetical documentation.</li> <li>Majority of Works Cited page is MLA correct.</li> <li>Most research info is Documented.</li> </ul>	<ul style="list-style-type: none"> <li>Inconsistent use of MLA style guidelines.</li> <li>Less than 3 cited sources used.</li> <li>Majority of parenthetical documentation done incorrectly.</li> <li>Random MLA documentation.</li> <li>Rarely documents sources.</li> </ul>	<ul style="list-style-type: none"> <li>Fails to follow MLA style Guidelines.</li> <li>Less than 3 cited sources used little or no parenthetical documentation.</li> <li>Works Cited page is not understandable.</li> </ul>
<b>Conventions</b>	<p>Superior editing. Rarely makes errors in the following areas:</p> <ul style="list-style-type: none"> <li>Spelling and mechanics.</li> <li>Correct usage and grammar.</li> </ul>	<p>Careful editing. Makes few errors in the following areas:</p> <ul style="list-style-type: none"> <li>Spelling and mechanics.</li> <li>Correct usage and grammar.</li> </ul>	<p>Some evidence of editing:</p> <ul style="list-style-type: none"> <li>Extensive spelling and grammatical errors.</li> </ul>	<p>Poor editing:</p> <ul style="list-style-type: none"> <li>Spelling and grammatical errors make it difficult to read paper.</li> </ul>

Score: \_\_\_\_\_ / 4

Evaluated by: \_\_\_\_\_ Date: \_\_\_\_\_

# Science Research Paper Rubric

**Barbara Houtz**

A red circular graphic with a gradient, appearing as a partial circle or a thick arc, located to the right of the author's name.

## Science Research Paper Rubric:

Rubrics for Assessing Student Achievement in Science Grades K-12 Hays B. Lantz, 2004-02-06 I recommend Rubrics for Assessing Student Achievement in Science Grades K 12 to any school district that is moving toward a standards based curriculum It will serve as a valuable tool for assessing student learning Grace Cisek Director of Mathematics and Science Curriculum Chester County Intermediate Unit PA At last science educators will now be able to use custom made rubrics to assess and evaluate student performance in the standards based science classroom Combining clarity detail utility and practicality veteran educator and author Hays B Lantz Jr offers the most complete collection of evaluation and assessment tools in science education available today This concise handbook was designed to improve the quality and uniformity of evaluation as well as assessment of student progress Written in language appropriate for both students and teachers in grades K 12 there are over 100 ready to use performance lists holistic rubrics and analytic rubrics that contain clear descriptions of the particular traits and qualities desired in student products and performances Key features distinguishing this book include Scoring tools for a wide range of products and performances found in effective science classrooms and programs Assessment tools that differentiate by learning levels providing a scaffolding of increasingly complex expectations across the grades Years of extensive field testing of the evaluative criteria Rubrics for Assessing Student Achievement in Science Grades K 12 is a valuable resource that will help to measure what students know and are able to do in the science classroom It will yield more consistent and defensible judgments more precise feedback and sharper student learning and performance China-us Science And Technology Competitiveness Assessment Report (2020) Debin Du, Dezhong Duan, 2021-04-23 Since 2018 the China US trade dispute has been escalating Science and technology S T and innovation are not only the underlying areas of trade dispute between the two largest economic powers but also the critical factors in determining whether China can win this trade war Notably after the ZTE event and Huawei event spotlight has been placed on the gap between the two countries in S T This book will introduce the competitiveness in S T a multifaceted concept and its evaluation index This is followed by a comparison of differences in S T development in China and the United States in the areas of human resources financial resources scientific research technological innovation and internationalization based on detailed and authoritative statistical data as well as in depth analysis of high tech industries such as the equipment manufacturing and information and communications technology *Helping Scientists to Communicate Well for All Considered: Strategic Science Communication in an Age of Environmental and Health Crises* Scott McWilliams, Marcia Allison, Marina Joubert, Ingrid Lofgren, Brianne Suldovsky, 2022-09-16 **Formative Assessment Strategies for Enhanced Learning in Science, K-8** Elizabeth Hammerman, 2008-09-18 Hammerman focuses on the meaning and centrality of formative assessment providing insights examples and ties to numerous individual initiatives to improve science experiences for students and to inspire better teaching for teachers Robert E Yager Professor of Science Education University of Iowa

Strategies for Teaching Science, Levels 6-12 Barbara Houtz, 2011-06-01 Developed for grades 6-12 this rich resource provides teachers with practical strategies to enhance science instruction. Strategies and model lessons are provided in each of the following overarching topics: inquiry and exploration, critical thinking and questioning, real-world applications, integrating the content areas and technology, and assessment. Research-based information and management techniques are also provided to support teachers as they implement the strategies within this resource. This resource supports core concepts of STEM instruction. **Science Framework for the 2011 National Assessment of Educational Progress, 2011**

<i>ASME Technical Papers</i> ,1976	<i>The Journal of Mental Science</i> ,1945	<b>Journal of Scientific &amp; Industrial Research</b> ,1987
U.S. Government Research & Development Reports ,1970	<b>Impact Assessment and Project Appraisal</b> ,2002	<u>Exceptional Child Education Resources</u> ,2002
<b>Resources in Education</b> ,1996	<b>PAIS Bulletin</b> ,1989	<i>Scientific American</i> ,1893
<b>Higher Education in the UK.</b> ,1995	<u>Proceedings of the Indiana Academy of Science</u> Indiana Academy of Science,1895	□□□□□□□□□□□□□□□□ □□□□□□ (Japan),1990
<b>Index of Conference Proceedings</b> British Library. Document Supply Centre,2001		

This book delves into Science Research Paper Rubric. Science Research Paper Rubric is a crucial topic that needs to be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Science Research Paper Rubric, encompassing both the fundamentals and more intricate discussions.

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    - Chapter 3: Science Research Paper Rubric in Everyday Life
    - Chapter 4: Science Research Paper Rubric in Specific Contexts
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  5. In chapter 4, the author will scrutinize the relevance of Science Research Paper Rubric in specific contexts. The fourth chapter will explore how Science Research Paper Rubric is applied in specialized fields, such as education, business, and technology.
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web jan 26 2016 the solubility product or ksp is a way of showing how effective a chemical is at dissolving into an aqueous solution the higher the ksp the more the chemical will dissolve into that solvent for this lab calcium hydroxide is created then ionized  $\text{Ca(OH)}_2$  when calcium is put in water reacts as follows  $\text{Ca} + 2\text{H}_2\text{O} \rightarrow \text{Ca(OH)}_2 + \text{H}_2$

*solubility product of calcium hydroxide chem lab* - Aug 07 2023

web k sp of calcium hydroxide 1 author j m mccormick last update october 13 2013 introduction an equilibrium constant  $K$  is related to  $\Delta G$  for a process through eqn 1 and as long as  $\Delta H$  and  $\Delta S$  are independent of temperature eqn 1 can be combined with eqn 2 to determine  $\Delta H$  and  $\Delta S$  from the temperature dependence of  $K$

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web sep 18 2020 in this experiment we will titrate a saturated calcium hydroxide solution with a standardized hcl solution we will determine the hydroxide ion concentration for the saturated calcium hydroxide

**determining the ksp of calcium hydroxide vernier** - Apr 03 2023

web your primary objective in this experiment is to test a saturated solution of calcium hydroxide and use your observations and measurements to calculate the ksp of the compound you will do this by titrating the prepared  $\text{Ca(OH)}_2$  solution with a standard hydrochloric acid solution

**chem 114 exp 9 mike university of rhode island** - May 04 2023

web in today's experiment you will determine the solubility product ksp of calcium hydroxide  $\text{Ca(OH)}_2$  by measuring the concentration of  $\text{Ca(OH)}_2$  in a saturated solution calcium hydroxide is a sparingly soluble salt that dissolves according to the following reaction the solubility product expression for this reaction is

*solved chem 182 determining the ksp of calcium hydroxide chegg* - Feb 18 2022

web expert verified chem 182 determining the ksp of calcium hydroxide calcium hydroxide is an ionic solid that is sparingly soluble in water a saturated aqueous solution of  $\text{Ca(OH)}_2$  is represented in equation form as shown below  $\text{Ca(OH)}_2(\text{s}) \rightleftharpoons \text{Ca}^{2+}(\text{aq}) + 2\text{OH}^{-}(\text{aq})$  the solubility product expression describes in mathematical terms the equilibrium

ksp for calcium hydroxide university of manitoba - Dec 31 2022

web ksp for calcium hydroxide calcium hydroxide is a soft white caustic powder used in making mortar cements calcium salts paints and petrochemicals it is also used in saltwater aquaria to make up kalkwasser limewater solutions for reef tanks and is used as a pH regulating agent notice that calcium hydroxide is divalent and thus has twice

*ksp calcium hydroxide chem132 determination of the ksp of calcium* - Jun 24 2022

web use the concentration of calcium hydroxide to calculate the concentration of calcium ions and hydroxide ions in the solution  $\text{Ca(OH)}_2 \rightarrow \text{Ca}^{2+} + 2\text{OH}^{-}$   $5 \times 10^{-2} \text{ M}$  calcium ions  $5 \times 10^{-2} \text{ M}$   $2 \times 10^{-2} \text{ M}$  hydroxide ions calculate calcium hydroxide ksp using the concentrations of calcium ions and hydroxide ions

**solubility of calcium hydroxide wpmu dev** - Oct 29 2022

web madison bowe elana keller april 15 2014 introduction the objective of this lab was to measure the ksp of a slightly soluble basic salt and how solubility changes with temperature ksp is another form of equilibrium expression that deals with solubility the reaction that will take place is  $\text{Ca(OH)}_2 \rightleftharpoons \text{Ca}^{2+} + 2\text{OH}^{-}$

**solved experiment 8 determining ksp for calcium hydroxide** - Aug 27 2022

web question experiment 8 determining ksp for calcium hydroxide post lab report sheet data table 2 points trial 1 trial 2 hci molarity 0.050 m volume of filtered  $\text{Ca(OH)}_2$  solution ml 5.00 5.00 3.71 13.75 equivalence point volume ml oh in saturated  $\text{Ca(OH)}_2$  m  $\text{Ca(OH)}_2$  in saturated  $\text{Ca(OH)}_2$  m ksp of  $\text{Ca(OH)}_2$  average ksp of  $\text{Ca(OH)}_2$

chemistry 1210 lab 11 experiment 11 determining the ksp of calcium - Sep 08 2023

web the chemical equation used was  $\text{Ca(OH)}_2 + 2\text{HCl} \rightarrow \text{CaCl}_2 + 2\text{H}_2\text{O}$  the solubility product was determined to be ksp  $\text{Ca(OH)}_2$  from the titration which was made by dripping a concentrated solution of calcium hydroxide into hydrochloric acid the calcium hydroxide was filtered before starting the titration

18.1 solubility product constant ksp chemistry libretexts - Feb 01 2023

web jul 12 2023 because the concentration of a pure solid such as  $\text{Ca}_3(\text{PO}_4)_2$  is a constant it does not appear explicitly in the equilibrium constant expression the equilibrium constant expression for the dissolution of calcium phosphate is therefore  $\text{Ca}_3(\text{PO}_4)_2 \rightleftharpoons 3\text{Ca}^{2+} + 2\text{PO}_4^{3-}$   $K_{\text{sp}} = [\text{Ca}^{2+}]^3 [\text{PO}_4^{3-}]^2$

**post lab report 8 determining the ksp of calcium hydroxide** - Jul 06 2023

web lab report 8 determining of calcium hydroxide the purpose of this was to titrate a saturated  $\text{Ca(OH)}_2$  solution with a standard hcl solution and find the ksp or the oh concentration was found and then the concentration was found and u for this experiment a titration of calcium hydroxide with hydrochloric

*solved determining the ksp of calciumhydroxide* - Apr 22 2022

web question determining the ksp of calciumhydroxide  $\text{Ca(OH)}_2 + 2\text{HCl} \rightarrow \text{CaCl}_2 + 2\text{H}_2\text{O}$  for this experiment the titration of calciumhydroxide with hydrochloric acid was investigated determining the ksp of calcium hydroxide  $\text{Ca(OH)}_2 + 2\text{HCl} \rightarrow \text{CaCl}_2 + 2\text{H}_2\text{O}$  for this experiment the titration of calcium

solved chem 182 experiment 8 determining the ksp of calcium chegg - Sep 27 2022

web question chem 182 experiment 8 determining the ksp of calcium hydroxide calcium hydroxide is an ionic solid that is sparingly soluble in water solution of  $\text{Ca(OH)}_2$  is represented in equation form  $\text{Ca(OH)}_2 \rightleftharpoons \text{Ca}^{2+} + 2\text{OH}^-$  lution of  $\text{Ca(OH)}_2$  on a nonionic solid that is sparingly soluble in water a saturated aqueous as shown below

determination of the ksp of calcium hydroxide autosaved - Jun 05 2023

web therefore it was determined through error calculations and ease of use that the most reliable valid method to use to get the closest ksp value to the theoretical ksp for  $\text{Ca(OH)}_2$  was the determination through ph method 2 introduction calcium hydroxide is a strong base soluble in water

exp 4 determining the solubility product constant ksp of calcium - Jul 26 2022

web plasma membrane homework symbiosis practice worksheet benzoin condensation grade a determining the ksp of

calcium hydroxide after reading the lab complete items a b c and d title purpose chemicals and equipment and summary of procedure as described on page 10 of exp 1 on an 8 1 2 x 11 sheet of paper

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web question lab 9 determination of the solubility product ksp of calcium hydroxide introduction ionic compounds that are classified as insoluble based on solubility rules are actually slightly solubla each of these insoluble compounds actually dissolves to

experimental determination of the solubility product of calcium hydroxide - Nov 29 2022

web we will filter the solid calcium hydroxide from the solution and then titrate the dissolved calcium hydroxide in solution with a known concentration of hydrochloric acid using titrimetric calculations we will determine the  $\text{Ca}^{2+}$  and  $\text{OH}^{-}$  concentrations in solution

**ksp of calcium hydroxide lab report for determining the ksp of** - Oct 09 2023

web calculate the ksp for calcium hydroxide  $\text{ksp Ca}^{2+} \text{OH}^{-}$  2 0 0 2  $\text{ksp } 5 \times 10^{-4}$  find the accepted value of the ksp for calcium hydroxide and compare it with your value discuss the discrepancy and suggest possible sources of experimental error accepted value for ksp for sodium hydroxide is  $5 \times 10^{-6}$

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