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web m x c x Δt equation 1 where m is the total mass of the solution solute plus solvent c is the specific heat of the solution and Δt is the observed temperature change the specific heat of the solution is generally assumed to be the same as that of

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web the heat of reaction Δh_{soln} is written after the products in units of kj mol rxn for example sodium hydroxide dissolves exothermically releasing 44.2 kilojoules per mole dissolved this process is represented as naoh s na aq oh aq Δh_{soln} 44.2 kj mol rxn write equations to similarly represent the dissolving process for

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