

2**MATTER AND CHANGE****SECTION 2.1 PROPERTIES OF MATTER (pages 39–42)**

This section helps you distinguish extensive from intensive properties and identify substances by their properties. It teaches you how to differentiate the three states of matter. It also defines a physical property and lists examples of physical properties and physical changes.

► Describing Matter (page 39)

1. The mass of an object is a measure of the amount of matter the object contains.
2. How does an extensive property differ from an intensive property?

An extensive property depends on the amount of matter, while an intensive property depends on the type of matter.

► Identifying Substances (page 40)

3. Matter that has a uniform and definite composition is called a substance.
4. Is the following sentence true or false? All samples of a substance have different physical properties. false
5. A physical property is a quality or condition of a substance that can be observed or measured without changing the substance's composition.
6. Circle the letter of the term that is NOT a physical property.
 - a. hardness
 - b. color
 - c. boiling point
 - d. melting
7. Look at Table 2.1 on page 40. What is the melting point of bromine? -7°C
8. Look at Table 2.1 on page 40. Circle the letter of the substance that is a yellow solid and melts at 115°C.
 - a. sulfur
 - b. chlorine
 - c. gold
 - d. copper

Section 3 1 Chemistry Answers

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Basic Concepts of Chemistry Leo J. Malone, Theodore O. Dolter, 2011-12-27 The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment. New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections, assessment exercises at the end of each section and relevant chapter problems at the end of each chapter. Every concept in the text is clearly illustrated with one or more step by step examples. Making it Real essays have been updated to present timely and engaging real world applications emphasizing the relevance of the material they are learning. This edition continues the end of chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter. WileyPLUS sold separately from text

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The Elements of Thermal Chemistry Matthew Moncrieff Pattison Muir, 1885

Micro-chemistry of Poisons Theodore George Wormley, 1885 **Russian Journal of General Chemistry**, 1998 **An**

Introduction to the Principles of Physical Chemistry from the Standpoint of Modern Atomistics and

Thermo-dynamics Edward Wight Washburn, 1915 **Russian Journal of Inorganic Chemistry**, 1987 *Journal of the Society of Chemical Industry* Society of Chemical Industry (Great Britain), 1898 Includes list of members 1882-1902 and proceedings of the annual meetings and various supplements

Report of the Annual Meeting British Association for the Advancement of Science, 1896 *Report of the ... Meeting of the British Association for the Advancement of Science* British Association for the Advancement of Science, 1896 *Indiana Pharmacist*, 1887 *The Lancet*, 1888 **Pharmaceutical**

Journal, 1896 **A Dictionary of Chemistry and the Allied Branches of Other Sciences** Henry Watts, 1879

Chemical News and Journal of Industrial Science, 1899 *A classified catalogue of ... educational works in use in Great Britain [by W. Low]*. Walter Low, 1871 **Catalogue of the educational division of the South Kensington**

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