



Weather Report - TG

Overview

The students are going on a virtual trip to another country. They are going to check the weather before they leave. The students will chart the weather in the country of their choice for today and check the forecast for the next three days. They will prepare a weather report, for television or radio, to perform in class.

Students can access all the Internet sites for the Interactive Activities by clicking on Students' Area in the ECB On-line homepage, then click on:

- Interactive Activity links - The links to the Internet sites are listed under the names of the activities.
- Or -
- Elementary, Junior High, High School. Choose their coursebook.

Level

Foundation and Intermediate (Grades 5-8)

Time

One 45-minute lesson

Group Size

Pairs or groups

Students' Prior Knowledge

The students ...

... can follow links on the Internet.

... can read a map.

... know directional vocabulary: north(N), south(S), east(E), west(W).

Objectives

The students will be able to ...

... identify different types of weather and their media symbols.

... report the differences in weather around the world using the Internet.

... give a weather forecast in the country chosen.

What Does A Weather Report Do

Shasha Hu



What Does A Weather Report Do:

Approximate Dynamic Programming Warren B. Powell, 2007-10-05 A complete and accessible introduction to the real world applications of approximate dynamic programming With the growing levels of sophistication in modern day operations it is vital for practitioners to understand how to approach model and solve complex industrial problems Approximate Dynamic Programming is a result of the author s decades of experience working in large industrial settings to develop practical and high quality solutions to problems that involve making decisions in the presence of uncertainty This groundbreaking book uniquely integrates four distinct disciplines Markov design processes mathematical programming simulation and statistics to demonstrate how to successfully model and solve a wide range of real life problems using the techniques of approximate dynamic programming ADP The reader is introduced to the three curses of dimensionality that impact complex problems and is also shown how the post decision state variable allows for the use of classical algorithmic strategies from operations research to treat complex stochastic optimization problems Designed as an introduction and assuming no prior training in dynamic programming of any form Approximate Dynamic Programming contains dozens of algorithms that are intended to serve as a starting point in the design of practical solutions for real problems The book provides detailed coverage of implementation challenges including modeling complex sequential decision processes under uncertainty identifying robust policies designing and estimating value function approximations choosing effective stepsize rules and resolving convergence issues With a focus on modeling and algorithms in conjunction with the language of mainstream operations research artificial intelligence and control theory Approximate Dynamic Programming Models complex high dimensional problems in a natural and practical way which draws on years of industrial projects Introduces and emphasizes the power of estimating a value function around the post decision state allowing solution algorithms to be broken down into three fundamental steps classical simulation classical optimization and classical statistics Presents a thorough discussion of recursive estimation including fundamental theory and a number of issues that arise in the development of practical algorithms Offers a variety of methods for approximating dynamic programs that have appeared in previous literature but that have never been presented in the coherent format of a book Motivated by examples from modern day operations research Approximate Dynamic Programming is an accessible introduction to dynamic modeling and is also a valuable guide for the development of high quality solutions to problems that exist in operations research and engineering The clear and precise presentation of the material makes this an appropriate text for advanced undergraduate and beginning graduate courses while also serving as a reference for researchers and practitioners A companion Web site is available for readers which includes additional exercises solutions to exercises and data sets to reinforce the book s main concepts

Monthly Weather Review ,1985 **Monthly Weather Review** United States. Weather Bureau,1887 Monthly Review
of the Iowa Weather and Crop Service ,1896 **Appendix to the Journals of the House of Representatives of New**

Zealand New Zealand. Parliament. House of Representatives, 1875 The American Meteorological Journal ,1894
English Mechanic and World of Science ,1897 **George Broomhall's Corn Trade News** ,1900 The Commercial & Financial Chronicle and Hunt's Merchants' Magazine ,1879 **"The" Athenaeum** ,1876 *Daily Weather Reports* ,1935
Report... Royal Society (Great Britain) Meteorological council, 1892 *Annual Report of the Board of Regents of the Smithsonian Institution* Smithsonian Institution. Board of Regents, 1885 Bulletin ,1892 *Nature London* ,1871
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What Does A Weather Report Do Introduction

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web jan 29 2020 the ionic form of this reaction is 2 na aq 2 cl aq cu 2 aq so 4 2 aq 2 na aq so 4 2 aq cucl 2 s the sodium ions and sulfate ion are the spectator ions in this reaction they appear unchanged in both the product and reactant side of the equation

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web mar 11 2014 solution ionic equation $\text{H}^+ \text{aq} + \text{Cl}^- \text{aq} + \text{Na}^+ \text{aq} + \text{HCO}_3^- \text{aq} + \text{Na}^+ \text{aq} + \text{Cl}^- \text{aq} + \text{H}_2\text{O} \text{l} + \text{CO}_2 \text{g}$ the spectator ions are Cl^- and Na^+
net ionic equation $\text{H}^+ \text{aq} + \text{Cl}^- \text{aq} + \text{Na}^+ \text{aq} + \text{HCO}_3^- \text{aq} + \text{Na}^+ \text{aq} + \text{Cl}^- \text{aq} + \text{H}_2\text{O} \text{l} + \text{CO}_2 \text{g} \rightarrow \text{H}^+ \text{aq} + \text{HCO}_3^- \text{aq} + \text{H}_2\text{O} \text{l} + \text{CO}_2 \text{g}$ here is a video on spectator ions
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web the six NO_3^- ions and the six Na^+ ions that appear on both sides of the equation are spectator ions that can be canceled to give the net ionic equation $3\text{Ba}^{2+} \text{aq} + 2\text{PO}_4^{3-} \text{aq} \rightarrow \text{Ba}_3(\text{PO}_4)_2 \text{s}$

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web these spectator ions whose presence is required to maintain charge neutrality are neither chemically nor physically changed by the process and so they may be eliminated from the equation to yield a more succinct representation called a

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 $\text{AgNO}_3 \text{aq} + \text{KCl} \text{aq} \rightarrow \text{AgCl} \text{s} + \text{KNO}_3 \text{aq}$ total ionic $\text{Ag}^+ \text{aq} + \text{NO}_3^- \text{aq} + \text{K}^+ \text{aq} + \text{Cl}^- \text{aq} \rightarrow \text{AgCl} \text{s} + \text{K}^+ \text{aq} + \text{NO}_3^- \text{aq}$ net ionic $\text{Ag}^+ \text{aq} + \text{Cl}^- \text{aq} \rightarrow \text{AgCl} \text{s}$ 2

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web video answer you know spectator ions are the ions present on the both sides of the equation indicate cancels out in the total ionic equations so you know in a case chemical reactions are described with chemical equations which symbolically present the starting materials end products

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web net ionic equations an equation for a reaction in solution showing only those particles that are directly involved in the chemical change spectator ions an ion that is not directly involved in a chemical reaction an ion that does not change oxidation number or composition during a reaction products

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web the equation below is the final net ionic equation after elimination of the spectator ions which yields water $\text{H}^+ \text{aq} + \text{OH}^- \text{aq} \rightarrow \text{H}_2\text{O} \text{l}$ the spectator ions in $\text{KOH} \text{aq}$ and $\text{HNO}_3 \text{aq}$ are $\text{K}^+ \text{aq}$ and $\text{NO}_3^- \text{aq}$ how to identify and list the

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web the ag aq and cl aq ions become agcl s but the na aq ions and the no 3 aq ions stay as na aq ions and no 3 aq ions these two ions are examples of spectator ions ions that do nothing in the overall course of a chemical reaction

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web aug 8 2022 a spectator ion is an ion that does not take part in the chemical reaction and is found in solution both before and after the reaction the net ionic equation is the chemical equation that shows only those elements compounds and ions that are directly involved in the chemical reaction

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