



## 04 Hyundai Terracan Manual

**Hyundai Motor Company (Seoul)**

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hot object is 60 c the specific heat of copper and water are 0 09 cal gm c and 1 cal gm c respectively

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web use the information in this chart to answer the following questions  $1$  if i have  $125 \text{ g}$  of water at a temperature of  $25.0^\circ\text{C}$  and heat it to  $145.0^\circ\text{C}$  what will the change in enthalpy be  $2$  sketch the heating curve temperature vs time for the process in  $1$   $3$  i have  $18.7 \text{ g}$  of an unknown liquid chemical at a temperature of  $35.0^\circ\text{C}$

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