

CONSTANTES, UNITÉS ET SYMBOLES USUELS

Quelques constantes fondamentales (IUPAC 2015)			
Grandeur physique	Symbolle	Valeur	Unités
constante d'Avogadro	N_A	6,02214179 · 10 ²³	mol^{-1}
vitesse de la lumière dans le vide	c, c_0	2,99792458 · 10 ⁸	$\text{m} \cdot \text{s}^{-1}$
accélération de la pesanteur	g	9,80665	$\text{m} \cdot \text{s}^{-2}$
charge élémentaire	e	1,602176487 · 10 ⁻¹⁹	C
volume molaire du gaz parfait à T.P.N. et sous 1 bar à 273,15 K	V_m	2,241410 · 10 ⁻²	$\text{m}^3 \cdot \text{mol}^{-1}$
constante des gaz	V_m^0	2,271108 · 10 ⁻²	$\text{m}^3 \cdot \text{mol}^{-1}$
	R	8,314472	$\text{J} \cdot \text{K}^{-1} \cdot \text{mol}^{-1}$
	R	0,08205783	$\text{L} \cdot \text{atm} \cdot \text{K}^{-1} \cdot \text{mol}^{-1}$
constante de Boltzmann	k, k_B	1,3806504 · 10 ⁻²³	$\text{J} \cdot \text{K}^{-1}$
constante de Planck	h	6,62606896 · 10 ⁻³⁴	J · s
constante de Faraday	F	9,64853399 · 10 ⁴	$\text{C} \cdot \text{mol}^{-1}$
constante de Rydberg	R_∞	1,0973731534 · 10 ⁷	m^{-1}
unité de masse atomique	m_u, u	1,6605402 · 10 ⁻²⁷	kg
masse de l'électron au repos	m_e	9,1093897 · 10 ⁻³¹	kg
masse du proton au repos	m_p	1,6726231 · 10 ⁻²⁷	kg
masse du neutron au repos	m_n	1,6749266 · 10 ⁻²⁷	kg
rayon de Bohr	a_0	5,29177249 · 10 ⁻¹¹	m
perméabilité du vide	μ_0	$4 \pi \cdot 10^{-7}$	$\text{N} \cdot \text{A}^{-2}$
permittivité du vide	ϵ_0	8,854187817 · 10 ⁻¹²	$\text{F} \cdot \text{m}^{-1}$

Alphabet grec					
Nom	minuscule	majuscule	Nom	minuscule	majuscule
alpha	α	Δ	nu	ν	Σ
béta	β	\Beta	xi	ξ	Ξ
gamma	γ	Γ	omicron	\o	\O
delta	δ	Δ	pi	π	Π
epsilon	ϵ	\Epsilon	rhô	ρ	\Rho
zéta	ζ	\Zeta	sigma	σ	Σ
éta	η	\Eta	tau	τ	\Tau
théta	θ	Θ	upsilon	υ	Υ
iota	ι	\Iota	phi	ϕ	Φ
kappa	κ	\Kappa	chi	χ	\Chi
lambda	λ	Λ	psi	ψ	Ψ
mu	μ	\Mu	oméga	ω	Ω

Tables De Constantes

K Morrison

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