

CONSTANTES, UNITÉS ET SYMBOLES USUELS

Quelques constantes fondamentales (IUPAC 2015)			
Grandeur physique	Symbole	Valeur	Unités
constante d'Avogadro	N_A	$6,02214179 \cdot 10^{23}$	mol^{-1}
vitesse de la lumière dans le vide	c, c_0	$2,99792458 \cdot 10^8$	$\text{m} \cdot \text{s}^{-1}$
accélération de la pesanteur	g	9,80665	$\text{m} \cdot \text{s}^{-2}$
charge élémentaire	e	$1,602176487 \cdot 10^{-19}$	C
volume molaire du gaz parfait à T.P.N.	V_m	$2,241410 \cdot 10^{-2}$	$\text{m}^3 \cdot \text{mol}^{-1}$
et sous 1 bar à 273,15 K	V_m°	$2,271108 \cdot 10^{-2}$	$\text{m}^3 \cdot \text{mol}^{-1}$
constante des gaz	R	8,314472	$\text{J} \cdot \text{K}^{-1} \cdot \text{mol}^{-1}$
	R	0,08205783	$\text{L} \cdot \text{atm} \cdot \text{K}^{-1} \cdot \text{mol}^{-1}$
constante de Boltzmann	k, k_B	$1,3806504 \cdot 10^{-23}$	$\text{J} \cdot \text{K}^{-1}$
constante de Planck	h	$6,62606896 \cdot 10^{-34}$	$\text{J} \cdot \text{s}$
constante de Faraday	F	$9,64853399 \cdot 10^4$	$\text{C} \cdot \text{mol}^{-1}$
constante de Rydberg	R_∞	$1,0973731534 \cdot 10^7$	m^{-1}
		$2,1798741 \cdot 10^{-18}$	J
unité de masse atomique	m_u, u	$1,6605402 \cdot 10^{-27}$	kg
masse de l'électron au repos	m_e	$9,1093897 \cdot 10^{-31}$	kg
masse du proton au repos	m_p	$1,6726231 \cdot 10^{-27}$	kg
masse du neutron au repos	m_n	$1,6749286 \cdot 10^{-27}$	kg
rayon de Bohr	a_0	$5,29177249 \cdot 10^{-11}$	m
perméabilité du vide	μ_0	$4\pi \cdot 10^{-7}$	$\text{N} \cdot \text{A}^{-2}$
permittivité du vide	ϵ_0	$8,854187817 \cdot 10^{-12}$	$\text{F} \cdot \text{m}^{-1}$

Alphabet grec					
Nom	minuscule	majuscule	Nom	minuscule	majuscule
Alpha	α	Λ	nu	ν	N
bêta	β	B	xi	ξ	Ξ
gamma	γ	Γ	omicron	\omicron	O
delta	δ	Δ	pi	π	Π
epsilon	ϵ	E	rho	ρ	P
zêta	ζ	Z	sigma	σ	Σ
êta	η	H	tau	τ	T
thêta	θ	Θ	upsilon	υ	Y
iota	ι	I	phi	ϕ	Φ
kappa	κ	K	chi	χ	X
lambda	λ	Λ	psi	ψ	Ψ
mu	μ	M	oméga	ω	Ω

Tables De Constantes

K Morrison



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