

Multiple Choice:

Selecting the alternative that constitutes the statement or answers the question. What is most commonly observed with water?

1. How many moles of hydroxide groups are a molecule of typical silicate tetrahedron?   
 a. 1   
 b. 2   
 c. 3   
 d. 4

2. For the reaction  $\text{Si}_2\text{O}_2 + 2\text{H}_2\text{O} \rightarrow 2\text{SiOH}$ , what factor determines whether a reaction?   
 a.  $\Delta G$    
 b.  $\Delta H$    
 c.  $\Delta S$    
 d.  $\Delta G^\circ$

3. The unhydrated silicate anion in the fused sil.   
 a.  $\text{SiO}_4^{4-}$    
 b.  $\text{SiO}_3^{2-}$    
 c.  $\text{SiO}_2$    
 d.  $\text{SiO}_4^{2-}$

4. For the reaction  $\text{Si}_2\text{O}_2 + 2\text{H}_2\text{O} \rightarrow 2\text{SiOH}$ ,  $\Delta G^\circ$  for the reaction is 0 kJ/mol. The reaction is spontaneous?   
 a. Yes   
 b. No   
 c. Not sure   
 d. Cannot tell

5. What is the study of the relationships between mineral products in natural systems?   
 a. Mineralogy   
 b. Petrology   
 c. Geochemistry   
 d. Mineralogy

6. What is the basic mineral silicate?   
 a.  $\text{SiO}_2$    
 b.  $\text{SiO}_4^{4-}$    
 c.  $\text{SiO}_2$    
 d.  $\text{SiO}_4^{2-}$

7. A mineral's empirical formula is  $\text{Si}_2\text{O}_2$ , at this ratio, silicon is 50 percent of the total weight of the mineral?   
 a. Yes   
 b. No   
 c. Not sure   
 d. Cannot tell

8. For the reaction  $\text{Si}_2\text{O}_2 + 2\text{H}_2\text{O} \rightarrow 2\text{SiOH}$ ,  $\Delta G^\circ$  for the reaction is 0 kJ/mol. The reaction is spontaneous?   
 a. Yes   
 b. No   
 c. Not sure   
 d. Cannot tell

9. For the reaction  $\text{Si}_2\text{O}_2 + 2\text{H}_2\text{O} \rightarrow 2\text{SiOH}$ ,  $\Delta G^\circ$  for the reaction is 0 kJ/mol. The reaction is spontaneous?   
 a. Yes   
 b. No   
 c. Not sure   
 d. Cannot tell

10. The chemical formula for quartz is  $\text{SiO}_2$ . What is the empirical formula?   
 a.  $\text{SiO}_2$    
 b.  $\text{SiO}_2$    
 c.  $\text{SiO}_2$    
 d.  $\text{SiO}_2$

11. The chemical formula for quartz is  $\text{SiO}_2$ . What is the empirical formula?   
 a.  $\text{SiO}_2$    
 b.  $\text{SiO}_2$    
 c.  $\text{SiO}_2$    
 d.  $\text{SiO}_2$

# Multiple Choice Stoichiometry Test

**Mary Millhollon, Richard H. Langley**



## Multiple Choice Stoichiometry Test:

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