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web 1 if the pe of the reactants is 430kj and the pe of the products is 190kj what is the  $\Delta H$  2 if the pe of the reactants is 430kj and the activation energy is 220 how much total energy will be needed to be added for the reaction to proceed transcribed image text potential energy diagram answer the questions using the graph

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$k = e^{-\frac{E_a}{RT}}$  where  $k$  depends on many factors as solvent geometry type of reaction  $E_a$  is the activation energy  $R$  is the universal constant of gases  $T$  is the absolute temperature

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