

## 14.3

## HYDROGEN IONS AND ACIDITY

## Section Review

## Objectives

1. Calculate concentrations based on the pH of acids and bases and vice versa.
2. Calculate pH and concentrations based on the  $K_a$  and  $K_b$  of acids and bases.
3. Calculate the pH of weak acids and bases.

## Vocabulary

- |                        |                             |                   |
|------------------------|-----------------------------|-------------------|
| 1. autoionization      | 2. conjugate acid-base pair | 3. weak acid/base |
| 4. autoionization      | 5. $K_a$                    | 6. $K_b$          |
| 7. ionization constant | 8. $K_a$                    | 9. $K_b$          |

## Key Equation

- |                                     |                                     |
|-------------------------------------|-------------------------------------|
| 1. $\text{pH} = -\log [\text{H}^+]$ | 2. $\text{pH} = -\log [\text{H}^+]$ |
| 3. $\text{pH} = -\log [\text{H}^+]$ | 4. $\text{pH} = -\log [\text{H}^+]$ |

## Part A Completion

Write the equation and/or the numerical value(s) of the  $K_a$  and  $K_b$  for the acids and bases listed below.

1.  $\text{HCl}$  is a strong acid.  $K_a = \text{_____}$

2.  $\text{HCl}$  is a strong acid.  $K_a = \text{_____}$

3.  $\text{HCl}$  is a strong acid.  $K_a = \text{_____}$

4.  $\text{HCl}$  is a strong acid.  $K_a = \text{_____}$

5.  $\text{HCl}$  is a strong acid.  $K_a = \text{_____}$

6.  $\text{HCl}$  is a strong acid.  $K_a = \text{_____}$

7.  $\text{HCl}$  is a strong acid.  $K_a = \text{_____}$

8.  $\text{HCl}$  is a strong acid.  $K_a = \text{_____}$

9.  $\text{HCl}$  is a strong acid.  $K_a = \text{_____}$

10.  $\text{HCl}$  is a strong acid.  $K_a = \text{_____}$

11.  $\text{HCl}$  is a strong acid.  $K_a = \text{_____}$

12.  $\text{HCl}$  is a strong acid.  $K_a = \text{_____}$

13.  $\text{HCl}$  is a strong acid.  $K_a = \text{_____}$

14.  $\text{HCl}$  is a strong acid.  $K_a = \text{_____}$

15.  $\text{HCl}$  is a strong acid.  $K_a = \text{_____}$

16.  $\text{HCl}$  is a strong acid.  $K_a = \text{_____}$

17.  $\text{HCl}$  is a strong acid.  $K_a = \text{_____}$

18.  $\text{HCl}$  is a strong acid.  $K_a = \text{_____}$

19.  $\text{HCl}$  is a strong acid.  $K_a = \text{_____}$

20.  $\text{HCl}$  is a strong acid.  $K_a = \text{_____}$

21.  $\text{HCl}$  is a strong acid.  $K_a = \text{_____}$

22.  $\text{HCl}$  is a strong acid.  $K_a = \text{_____}$

23.  $\text{HCl}$  is a strong acid.  $K_a = \text{_____}$

# 19 2 Hydrogen Ions And Acidity Worksheet Answers

**Camilla Rothe**



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