

**ECONOMICS
GRADE 10
CLASS TEST**

**MARKS: 40
TIME : 1 HOUR**

**SECTION A
QUESTION 1**

1.1. Choose the correct answer and write only the letter (A-D) next to the question number (1.1.1. – 1.1.5)

1.1.1. Which of the following is NOT an oligopoly market?

- a. MTN, Vodacom, and Cell C
- b. FNB, ABSA and Standard Bank
- c. Engen, Caltex and Sasol
- d. KFC, Nandos and Hungry Lion

2

1.1.2. The maximum amount of money a consumer is willing to sacrifice for a good or a service is known as ...

- a. Satisfaction
- b. Value
- c. Price
- d. A market

2

1.1.3. World markets are also known as markets.

- a. Collective
- b. Money
- c. Global
- d. Perfect

2

1.1.4. A characteristic of an imperfect market:

- a. There is one supplier or a few large suppliers
- b. Products are homogenous
- c. There are no barriers to entry
- d. There are many buyers and sellers

2

1.1.5. Buyers and sellers meet in a ... to determine prices

- a. Foreign Sector
- b. Market
- c. Bank
- d. Supply

2

2 x 5 = 10

Economics March Common Test Question Paper And Memorandum 2014

David Kirk



Economics March Common Test Question Paper And Memorandum 2014:

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Semiannual with semiannual and annual indexes References to all scientific and technical literature coming from DOE its laboratories energy centers and contractors Includes all works deriving from DOE other related government sponsored information and foreign nonnuclear information Arranged under 39 categories e g Biomedical sciences basic studies Biomedical sciences applied studies Health and safety and Fusion energy Entry gives bibliographical information and abstract Corporate author subject report number indexes **Far Eastern Economic Review** ,1956 **Bibliography of Free and Inexpensive Materials for Economic Education** Joint Council on Economic Education,1955 **Business and Economic Studies - Bureau of Business Research, Dept. of Economics and Business Administration, West Virginia University** ,1949 *Streaming and Copyright Law* Lasantha Ariyaratna,2022-09-05 This book examines the challenges posed to Australian copyright law by streaming from the end user perspective It compares the Australian position with the European Union and United States to draw lessons from them regarding how they have dealt with streaming and copyright By critically examining the technological functionality of streaming and the failure of copyright enforcement against the masses it argues for strengthening end user rights The rising popularity of streaming has resulted in a revolutionary change to how digital content such as sound recordings cinematographic films and radio and television broadcasts is used on the internet Superseding the conventional method of downloading using streaming to access digital content has challenged copyright law because it is not clear whether end user acts of streaming constitute copyright infringement These prevailing grey areas between copyright and streaming often make end users feel doubtful about accessing digital content through streaming It is uncertain whether exercising the right of reproduction is appropriately suited for streaming given the ambiguities of embodiment and scope of substantial part Conversely the fair dealing defence in Australia cannot be used aptly to defend end users acts of streaming digital content because end users who use streaming to access digital content can rarely rely on the defence of fair dealing for the purposes of criticism or review news reporting parody or satire or research or study When considering a temporary copy exception end users are at risk of being held liable for infringement when using streaming to access a website that contains infringing digital content even if they lack any knowledge about the content s infringing nature Moreover the grey areas in circumventing geo blocking have made end users hesitant to access websites through streaming because it is not clear whether technological protection measures apply to geo blocking End users have a severe lack of knowledge about whether they can use circumvention methods such as virtual private networks

to access streaming websites without being held liable for copyright infringement Despite the intricacies between copyright and access to digital content the recently implemented website blocking laws have emboldened copyright owners while suppressing end users access to digital content This is because the principles of proportionality and public interest have been given less attention when determining website blocking injunctions Economics Working Papers: a Bibliography ,1985

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for the nitration 4 01 ml of nitric acid 4 00 ml of sulfuric acid and 4 56 ml bromobenzene were used to complete the reaction with these measurements bromobenzene was found to be the limiting reagent since only 4 56 ml of bromobenzene was used the amount of product predicted to be recovered was 8 7533 g

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in this experiment we will carry out the nitration of bromobenzene bromine substituted on an aromatic ring is an ortho para directing group it will direct an incoming electrophile to the ortho and para positions it is also deactivating meaning that its reactions will be slower than the identical reactions of benzene

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section 20711 lab experiment nitration of bromobenzene introduction the chemical reaction of nitration consists of a nitro group being added to or substituted in a molecule nitration can basically be carried out by a mixture of concentrated nitric

acid and sulfuric acid this mixture is useful to obtain the active nitronium ion

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1 go to the restroom and wash the affected area with lots of soap and water the soap in the lab is not suitable for this purpose 2 return to the lab and apply mineral oil to the affected area 3 the summary to this warning is not to touch any part of your body while performing this experiment if you must leave the lab for any reason 1

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in the nitration of bromobenzene a mixture of sulfuric acid and nitric acid react with one another nitric acid the weaker of the two acids will serve as base to form nitronium ion bromobenzene will attack the nitronium ion by nucleophilic attack

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nitration of bromobenzene lab report improved essays 731 words 3 pages open document essay sample check writing quality show more this experiment was performed in order to demonstrate the principles of nitration through electrophilic aromatic substitution it also examined the effects of bromine on an aromatic ring ortho para vs meso directing

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then actual is your actual findings yield actual theoretical $\times 100$ what are the possible side reactions of nitration of bromobenzene dinitration can occur at ortho para or at both ortho positions occurs in overheating meta product happens very rarely because bromine is an ortho para director

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the soap in the lab is not suitable for this purpose 2 return to the lab and apply mineral oil to the affected area theory and explanation of the nitration of bromobenzene if the temperature of the reaction mixture exceeds 60 °C

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the bromine substituent in bromobenzene is an ortho para director the para position is favored due to steric effects in the reaction that takes place in this lab a nitro group is substituted onto bromobenzene at the para position to form 1-bromo-4-nitrobenzene

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study with quizlet and memorize flashcards containing terms like reacting sulfuric and nitric acid together produces the nitronium ion alone nitric acid is not a strong enough electrophile to react with bromobenzene sulfuric acid is added to the nitric acid the reverse rxn can overheat and boil easily includes a claisen adaptor and

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over 5 minutes slowly pipet this bromobenzene into the icy cold nitration mixture HNO_3 3 h 2 so 4 in the 125 ml flask and stir continue to stir the reaction mixture in the ice bath for 30 minutes do not heat in a new 250 ml erlenmeyer

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on the right the arenium cation has a nitro set with incredibly small 0.6 angstrom bond distances this will result in an optimization failure as the atoms are even close for the programme to accomodate their electrons demonstration of creating the p arenium cation by to nitration of bromobenzene

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the purpose of this lab was to observe where substituents are added on a benzene ring by way of electrophilic aromatic substitution the electrophilic source in the nitration of bromobenzene comes from a nitronium ion NO_2^+ which acts as the electrophilic species

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nitration of bromobenzene is an electrophilic aromatic substitution reaction the electrophile will be formed from 4.0ml of nitric acid and 4.0ml of sulfuric acid and will then react with 4.5ml of bromobenzene this will produce o bromonitrobenzene and p bromonitrobenzene

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lab report 10 nitration of bromobenzene raekwon filmore cm 244 section 40 march 27 2018 introduction for this experiment nitration of bromobenzene was the focus of the lab the benzene is an aromatic compound and when it reacts with wither a mixture of sulfuric acid or nitric acid creates what is known as a nitro group

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