



THE UNIVERSITY OF PAPUA NEW GUINEA

UPNG EXTERNAL APPLICATION FORM FOR 2026



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SECTION I: SUMMARY OF CHOICES

This application is for (Tick a Box):

☐

New Enrolment for External Program to:

- ☐ School of Business and Public Policy Management
- ☐ School of Humanities & Social Sciences
- ☐ School of Natural & Physical Sciences

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George Nicholas, Joe Watkins

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buffers typically consist of an acid base pair with the acid and base differing by the presence or absence of a proton a conjugate acid base pair for instance one of the buffers that maintain the pH of human blood involves carbonic acid H_2CO_3 and its conjugate base the bicarbonate ion HCO_3^-

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acid base titration difference between acid and base acids and bases definition an acid is any hydrogen containing substance that is capable of donating a proton hydrogen ion to another substance a base is a molecule or ion able to accept a hydrogen ion from an acid acidic substances are usually identified by their sour taste

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the reaction of an acid and a base is called a neutralization reaction although acids and bases have their own unique chemistries the acid and base cancel each other's chemistry to produce a rather innocuous substance water in fact the general reaction between an acid and a base is

$$\text{acid} + \text{base} \rightarrow \text{water} + \text{salt}$$

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