

Multiple Choice

Select the alternative that best completes the statement or answers the question. What is the most common primary gas used?

1. How many moles of hydrogen gas are in a cylinder at 100 atm pressure, 200°C?
- ☐ 1  
☒ 2  
☐ 3  
☐ 4

2. At constant  $T_1 = 100^\circ\text{C}$ , what is the molar volume of an ideal gas?
- ☒ 22.4  
☐ 22.7  
☐ 22.8  
☐ 22.9

3. The universal gas constant is the same as:
- ☐ Boltzmann's constant  
☐ gas volume  
☒ heat capacity  
☒ work function

4. For a constant  $T_1 = 100^\circ\text{C}$ ,  $n = 1.0$  mol, how many moles of gas are in a cylinder at 100 atm pressure, 200°C?
- ☐ 1.0 mol  
☐ 2.0 mol  
☐ 3.0 mol  
☒ 4.0 mol

5. What is the value of the universal gas constant?
- ☐ 8.314 J/mol·K  
☒ 8.314 J/mol·K  
☐ 8.314 J/mol·K  
☐ 8.314 J/mol·K

6. What is the value of the universal gas constant?
- ☐ 8.314 J/mol·K  
☐ 8.314 J/mol·K  
☒ 8.314 J/mol·K  
☐ 8.314 J/mol·K

7. A constant volume calorimeter is used to determine the heat of combustion of a substance. The heat of combustion is:
- ☒ 22.4  
☐ 22.7  
☐ 22.8  
☐ 22.9

8. For a constant  $T_1 = 100^\circ\text{C}$ ,  $n = 1.0$  mol, how many moles of gas are in a cylinder at 100 atm pressure, 200°C?
- ☒ 1.0 mol  
☐ 2.0 mol  
☐ 3.0 mol  
☐ 4.0 mol

9. At constant  $T_1 = 100^\circ\text{C}$ , what is the molar volume of an ideal gas?
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☐ 22.9

10. The universal gas constant is the same as:
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☒ heat capacity  
☐ work function

11. The universal gas constant is the same as:
- ☐ Boltzmann's constant  
☐ gas volume  
☒ heat capacity  
☐ work function

# Multiple Choice Stoichiometry Test

**David Wilson**



## Multiple Choice Stoichiometry Test:

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