

14.3

HYDROGEN IONS AND ACIDITY

Section Review

Objectives

1. Calculate concentrations based on the pH of acids and bases and vice versa.
2. Calculate pH and concentrations based on K_a and K_b and vice versa.
3. Calculate the pH of weak acids and bases.

Vocabulary

- | | | |
|-------------------|-----------------------------|---------------------|
| 1. autoionization | 2. conjugate acid-base pair | 3. acid solution |
| 4. autoionization | 5. acid rain | 6. neutral solution |
| | 7. acid solution | 8. pH |

Key Equation

- | | |
|-------------------------------------|-------------------------------------|
| 1. $\text{pH} = -\log [\text{H}^+]$ | 2. $\text{pH} = -\log [\text{H}^+]$ |
| 3. $\text{pH} = -\log [\text{H}^+]$ | 4. $\text{pH} = -\log [\text{H}^+]$ |

Part A Completion

Write the equation and/or the numerical value(s) for the reaction(s) indicated. Show the calculation for pH or pOH. Write the chemical formula for the acid or base indicated.

1. What substance is _____? Write the chemical formula.
2. Write the chemical formula for _____? Write the chemical formula of the acid or base.
3. Write the chemical formula for _____? Write the chemical formula.
4. Write the chemical formula for _____? Write the chemical formula.
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18. Write the chemical formula for _____? Write the chemical formula.
19. Write the chemical formula for _____? Write the chemical formula.
20. Write the chemical formula for _____? Write the chemical formula.

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19 2 Hydrogen Ions And Acidity Work Sheet Answers

M Tight



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