

## Section 1: Introduction

Because of the high concentration of the product, the reaction mixture was extremely viscous. The reaction mixture with the addition of the starting ether dissolved in a solution of  $\text{LiClO}_4$  and  $\text{CH}_3\text{CO}_2\text{Et}$  was then deoxygenated. After a 10-min. period, the reaction mixture was heated to 100°. The dry ice was then added and the ether solution was collected. The product was purified by column chromatography on alumina. The product was a colorless, viscous oil. The infrared spectrum was identical with that of the product that resulted from the reaction of the starting ether with  $\text{CH}_3\text{CO}_2\text{Et}$  and  $\text{LiClO}_4$  in  $\text{CH}_3\text{CO}_2\text{Et}$  at 100°.

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• **Geometric mean of the sample** =  $\sqrt[n]{\prod_{i=1}^n x_i}$  =  $\sqrt[n]{\prod_{i=1}^n (a_i + b_i)}$  =  $\sqrt[n]{\prod_{i=1}^n a_i + \prod_{i=1}^n b_i}$

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