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Léo Pomar



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web 1 $3\text{NaBr} + \text{H}_3\text{PO}_4 \rightarrow \text{Na}_3\text{PO}_4 + 3\text{HBr}$ type of reaction double displacement 2 $3\text{Ca(OH)}_2 + \text{Al}_2\text{SO}_4 \rightarrow 3\text{CaSO}_4 + 2\text{Al(OH)}_3$ type of reaction double displacement 3 $3\text{Mg} + \text{Fe}_2\text{O}_3 \rightarrow 3\text{MgO} + 2\text{Fe}$ type of reaction single displacement 4 $\text{C}_2\text{H}_4 + 3\text{O}_2 \rightarrow 2\text{CO}_2 + 2\text{H}_2\text{O}$ type of reaction combustion 5 $2\text{PbSO}_4 + 2\text{PbSO}_3 + \text{O}_2$ type of reaction

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