

WW2 topic quiz

Page 1 of 3

EASIER	(SOME) HARDER
Question 1: What date did Britain declare war on Germany?	
A. 3 rd September 1938 B. 3 rd September 1939 C. 3 rd September 2039	A. 13 th September 1939 B. 3 rd September 1939 C. 1 st September 1939
Question 2: What date did the war in Europe end?	
A. 8 th May 1944 B. 8 th May 1945 C. 8 th May 1946	A. 8 th May 1945 B. 9 th May 1945 C. 10 th May 1945
Question 3: What year did the USA join the war?	
A. 1940 B. 1941 C. 1942	
Question 4: Which one of these statements is correct?	
A. Winston Churchill was the Chancellor of Germany B. Winston Churchill was the Prime Minister of Britain C. Winston Churchill was in a boy band called "WW2 Direction"	A. Winston Churchill was replaced by Neville Chamberlain B. Neville Chamberlain was replaced by Winston Churchill C. Winston Churchill was in a boy band called "WW2 Direction"
Question 5: Which one of these statements is correct?	
A. Albert Hitler was the German leader B. Rudolph Hitler was the German leader C. Adolf Hitler was the German leader	A. Adolf Hitler came to power in 1933 and invaded Poland in 1941 B. Rudolph Hitler came to power in 1933 and invaded Poland in 1939 C. Adolf Hitler came to power in 1933 and invaded Poland in 1939
Question 6: What was the proper army name for the bombs that burst into flames on landing?	
A. Incendiary bombs B. Incendiary bombs C. Firebomb Fritz	
Question 7: Why was the V2 so deadly compared to the V1?	
A. It moved faster than sound so people couldn't hear it coming B. It was thinner so it was harder to see C. It was much more thinner	
Question 8: Why were children evacuated?	
A. To give their parents some peace B. For a nice holiday C. To keep them safe	
Question 9: Why were lampposts and pavements painted with white stripes?	
A. So that people knew where to park in the dark B. So that people could see where they were going in the dark C. So that enemy zebras could hide	
Question 10: What was the popular phrase that the air raid wardens used?	
A. Put that cat out B. Put that light out C. Put your left leg out and your left leg in	

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web we will revisit this today and go on to see that if we use a special kind of thermochemical reaction called the standard enthalpy of formation we can calculate enthalpies of reactions without having to manipulate a series of individual thermochemical equations for each step

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web we can express the first law in many ways one of the more useful expressions is that the change in internal energy ΔE of a system in any process is equal to the heat Q added to the system plus the work W done on the system by its surroundings $\Delta E = Q + W$

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