

Multiple Choice:

Selecting the alternative that constitutes the statement or answers the question. What is most commonly observed with water?

1. How many moles of hydrogen gas are in a cylinder at typical conditions, 1000 L?
 a. 1
 b. 10
 c. 100
 d. 1000

2. At temperature $T_1 = 100^\circ\text{C}$, water has a vapor pressure of 1000 mmHg.
 a. 100
 b. 10
 c. 1
 d. 0.1

3. The enthalpy of fusion is the amount of:
 a. heat of the melting
 b. heat of the solidification
 c. heat of the vaporization
 d. heat of the condensation

4. For temperature $T_1 = 100^\circ\text{C}$, $\Delta H_{\text{vap}} = 40.7 \text{ kJ/mol}$, how many moles of water vapor are produced when 1000 J of heat is added?
 a. 10 moles
 b. 1.0 mole
 c. 0.1 mole
 d. 0.01 mole

5. What is the study of the intermolecular forces between molecules in condensed phases?
 a. chemical thermodynamics
 b. statistical thermodynamics
 c. physical chemistry
 d. molecular thermodynamics

6. What is the boiling point of water at 1 atm?
 a. 100°C
 b. 100 K
 c. 373°C
 d. 373 K

7. A substance's empirical formula is CH_2 , at this ratio, what is its molar mass?
 a. 14 g/mol
 b. 28 g/mol
 c. 42 g/mol
 d. 56 g/mol

8. For temperature $T_1 = 100^\circ\text{C}$, $\Delta H_{\text{vap}} = 40.7 \text{ kJ/mol}$, how many moles of water vapor are produced when 1000 J of heat is added?
 a. 10 moles
 b. 1.0 mole
 c. 0.1 mole
 d. 0.01 mole

9. At temperature $T_1 = 100^\circ\text{C}$, $\Delta H_{\text{vap}} = 40.7 \text{ kJ/mol}$, how many moles of water vapor are produced when 1000 J of heat is added?
 a. 10
 b. 1
 c. 0.1
 d. 0.01

10. What is the molar mass of water?
 a. 18 g/mol
 b. 36 g/mol
 c. 72 g/mol
 d. 144 g/mol

11. The molar mass of water is 18 g/mol. How many moles of water are in 1000 g of water?

Multiple Choice Stoichiometry Test

Mary Millhollon, Richard H. Langley



Multiple Choice Stoichiometry Test:

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