

19.2

HYDROGEN IONS AND ACIDITY

Section Review

Objectives

1. Calculate K_a and K_b for weak acids and bases, and calculate pK_a and pK_b for weak acids and bases.
2. Calculate K_a and K_b for weak acids and bases, and calculate pK_a and pK_b for weak acids and bases.
3. Calculate the pH of weak acid and base solutions.

Vocabulary

- | | | |
|-----------------|--------------------------------------|------------------------------------|
| acid | acid dissociation constant (K_a) | acid ionization constant (K_a) |
| acidic solution | acidic solution | acidic solution |
| acidic solution | acidic solution | acidic solution |

Key Equations

- | | |
|---|--|
| $K_a = \frac{[H^+][A^-]}{[HA]}$ (Equation 19.1) | $K_b = \frac{[OH^-][HB^+]}{[B]}$ (Equation 19.2) |
|---|--|

Part A Completion

Write the K_a expression for the following acids. Calculate K_a and pK_a for the following acids. Calculate K_b and pK_b for the following bases. Calculate pH and pOH for the following solutions.

1. HCl (strong acid) $K_a = \text{_____}$ $pK_a = \text{_____}$

2. HNO_3 (strong acid) $K_a = \text{_____}$ $pK_a = \text{_____}$

3. H_2SO_4 (strong acid) $K_a = \text{_____}$ $pK_a = \text{_____}$

4. H_2SO_3 (weak acid) $K_a = \text{_____}$ $pK_a = \text{_____}$

5. H_2CO_3 (weak acid) $K_a = \text{_____}$ $pK_a = \text{_____}$

6. $H_2PO_4^-$ (weak acid) $K_a = \text{_____}$ $pK_a = \text{_____}$

7. $H_2PO_4^-$ (weak acid) $K_a = \text{_____}$ $pK_a = \text{_____}$

8. $H_2PO_4^-$ (weak acid) $K_a = \text{_____}$ $pK_a = \text{_____}$

9. $H_2PO_4^-$ (weak acid) $K_a = \text{_____}$ $pK_a = \text{_____}$

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12. $H_2PO_4^-$ (weak acid) $K_a = \text{_____}$ $pK_a = \text{_____}$

13. $H_2PO_4^-$ (weak acid) $K_a = \text{_____}$ $pK_a = \text{_____}$

14. $H_2PO_4^-$ (weak acid) $K_a = \text{_____}$ $pK_a = \text{_____}$

15. $H_2PO_4^-$ (weak acid) $K_a = \text{_____}$ $pK_a = \text{_____}$

16. $H_2PO_4^-$ (weak acid) $K_a = \text{_____}$ $pK_a = \text{_____}$

17. $H_2PO_4^-$ (weak acid) $K_a = \text{_____}$ $pK_a = \text{_____}$

18. $H_2PO_4^-$ (weak acid) $K_a = \text{_____}$ $pK_a = \text{_____}$

19. $H_2PO_4^-$ (weak acid) $K_a = \text{_____}$ $pK_a = \text{_____}$

20. $H_2PO_4^-$ (weak acid) $K_a = \text{_____}$ $pK_a = \text{_____}$

21. $H_2PO_4^-$ (weak acid) $K_a = \text{_____}$ $pK_a = \text{_____}$

22. $H_2PO_4^-$ (weak acid) $K_a = \text{_____}$ $pK_a = \text{_____}$

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Yan Bai



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