

Semester 2 Final Exam Review – Answer Key

1. Mendeleev
2. Electrons in the outer-most energy shell of an atom
3. F, O, N
4. -ide
5. Atoms with a full valence shell
6. Molarity (M) = moles per liter = moles/liters
7. Basic, alkaline
8. Group or Family
9. $6.02 \times 10^{23} \times 5 = 3.01 \times 10^{24}$
10. Fluid, compressible, take the shape and volume of the container, molecules are far apart (low density)
11. Period
12. Acidic
13. High boiling point, low vapor pressure
14. A homogeneous mixture of only one phase; solute dissolved in a solvent
15. 22.4L
16. Endothermic
17. $\text{Zn}(\text{NO}_3)_2$
18. H, N, O, F, Cl, Br, I
19. Nature of the reactants, temperature, surface area, concentration, catalyst
20. Calorie = energy required to raise 1g of water 1 degree Celcius
21. Volume is directly proportional to temperature @ a constant pressure ($V_1/T_1 = V_2/T_2$)
22. Iron(II) chloride
23. $\text{Ba}(\text{NO}_3)_2$
24. PbCO_3
25. Reactants react to form products
26. Synthesis, decomposition, combustion, single replacement, double replacement
27. Water and carbon dioxide
28. Right side, $-\Delta H$
29. The energy require to form an activated complex and have a reaction proceed to form products
30. Exothermic, $\Delta H = -585\text{kJ}$
 - Convert mass to moles: $34\text{g} + 17\text{g/mol} = 2 \text{ mol NH}_3$
 - Multiply by energy/mole ratio: $2 \text{ mol NH}_3 \times (-1170\text{kJ})/4 \text{ mol NH}_3 = -585\text{kJ}$
31. Concentration, temperature, pressure (gas, heterogeneous system)
32. A buffer solution is a solution of a conjugate acid-base pair. A buffer system prevents significant changes in pH when a small amount of acid or base is added to the solution. Your body has a bicarbonate buffer system in the blood which helps maintain you blood's pH.
33. Unsaturated = more room for solute @ temp; saturated = solute maxed out @ temp; supersaturated = more solute than maximum for given temp

Nd Semester Final Exam Review Science Answer

Logan Fiorella, Richard E. Mayer



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Learning as a Generative Activity Logan Fiorella, Richard E. Mayer, 2015-02-05 This book presents eight evidence based strategies that promote generative learning which enables learners to apply their knowledge to new problems *Annual Catalog - United States Air Force Academy* United States Air Force Academy, 1971 United States Air Force Academy United States Air Force Academy, 1972 **Annual Catalogue** United States Air Force Academy, 1984 **Unforgettable** W. Michael Gray, 2016-10-12 We have an uneasy relationship with the relentless deluge of information gushing out of academia and our media outlets To turn it off is escapist but to attempt to cognitively grapple with it is overwhelming In *Unforgettable* Enabling Deep and Durable Learning a nationally recognized master teacher gives professors and their students the means to chart a clear path through this information explosion Humans crave explanatory patterns and this book enables teachers to think deeply about their academic disciplines to find and articulate their core explanatory principles and to engage their students in a compelling way of thinking An alternative title for this book could be *Why the Best College Teachers Do What They Do* because the author articulates a compelling rationale that will equip faculty to create and deliver transformative courses Students in transformative courses grapple with essential questions and gain mental muscle that equips them for real world challenges **The Science Teacher**, 1968 SCC Library has 1964 cur **Official Report, of the Student Curriculum Evaluation Committee, 1968-1969** University of Virginia. College of Arts and Sciences. Student Curriculum Evaluation Committee, 1969 **Biology** Edward C. Gruber, 1963 **Public Policy** Allan Kornberg, 1989 Cornell University Courses of Study Cornell University, 2006 Women in Engineering Conference, 1991 **The 2002 Guide to the Evaluation of Educational Experiences in the Armed Services** American Council on Education, 2002 Long considered to be the standard reference work in this area this three volume set describes more than 8 000 courses offered between January 1990 and the present by various service branches and the Department of Defense Long considered to be the standard reference work in this area this three volume set describes more than 8 000 courses offered between January 1990 and the present by various service branches and the Department of Defense Updated every two years **Certification and Accreditation Programs Directory** Michael A. Paré, 1996 Providing detailed profiles on certification and accreditation programmes in the US this book includes information on certification and accreditation programmes that denote skill level professionalism accomplishment and excellence MAA Notes, 1983 *The National Guide to Educational Credit for Training Programs* American Council on Education, 2005 Highlights over 6 000 educational programs offered by business labor unions schools training suppliers professional and voluntary associations and government agencies Journal of Engineering Education, 2001 **Catalog** University of Colorado Boulder, 2005 Tests in Print, 1994 **Tests in Print IV** Linda L. Murphy, Jane Close Conoley, James C. Impara, 1994 Customers who place a standing order for the Tests in Print series or the Mental Measurements Yearbook series will receive a 10% discount on every volume To place your standing

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