

Chemistry Study Guide

Periodic Table of the Elements

The periodic table displays elements organized by atomic number, with groups labeled 1 through 10 and periods labeled 1 through 7. Lanthanide and actinide series are shown at the bottom.

Objective:

To understand the *structure of an atom*, and the *nature + properties* of subatomic particles ✎📌

Matter:

Anything that has **mass** and **occupies space** ✎📌

Examples: air, gases, liquids, solids!

Atoms:

Atoms are the **smallest, indivisible particle** ✎📌

- **Outer part** → *shell/orbit/energy level* → carries the electrons (e^-)
- **Inner part** → *nucleus* → houses the protons (p^+) and neutrons (n)
- Outer & Inner part → together these are the **subatomic particles**

Subatomic Particle Table:

Particle	Symbol	Mass (amu)	Charge	Location
Proton	p^+	1	+1	Nucleus
Neutron	n	1	0	Nucleus
Electron	e^-	~ 0	-1	Orbit/Shell

Section 2 nuclear Chemistry Study Guide Key

Valentine Korah



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