

Semester 2 Final Exam Review – Answer Key

1. Mendeleev
2. Electrons in the outer-most energy shell of an atom
3. F, O, N
4. -ide
5. Atoms with a full valence shell
6. Molarity (M) = moles per liter = moles/liters
7. Basic, alkaline
8. Group or Family
9. $6.02 \times 10^{23} \times 5 = 3.01 \times 10^{24}$
10. Fluid, compressible, take the shape and volume of the container, molecules are far apart (low density)
11. Period
12. Acidic
13. High boiling point, low vapor pressure
14. A homogeneous mixture of only one phase; solute dissolved in a solvent
15. 22.4L
16. Endothermic
17. $Zn(NO_3)_2$
18. H, N, O, F, Cl, Br, I
19. Nature of the reactants, temperature, surface area, concentration, catalyst
20. Calorie = energy required to raise 1g of water 1 degree Celsius
21. Volume is directly proportional to temperature @ a constant pressure ($V_1/T_1 = V_2/T_2$)
22. Iron(II) chloride
23. $Ba(NO_3)_2$
24. $PbCO_3$
25. Reactants react to form products
26. Synthesis, decomposition, combustion, single replacement, double replacement
27. Water and carbon dioxide
28. Right side, $-\Delta H$
29. The energy require to form an activated complex and have a reaction proceed to form products
30. Exothermic, $\Delta H = -585\text{kJ}$
 - Convert mass to moles: $34\text{g} + 17\text{g/mol} = 2 \text{ mol NH}_3$
 - Multiply by energy/mole ratio: $2 \text{ mol NH}_3 \times (-1170\text{kJ})/4 \text{ mol NH}_3 = -585\text{kJ}$
31. Concentration, temperature, pressure (gas, heterogeneous system)
32. A buffer solution is a solution of a conjugate acid-base pair. A buffer system prevents significant changes in pH when a small amount of acid or base is added to the solution. Your body has a bicarbonate buffer system in the blood which helps maintain your blood's pH.
33. Unsaturated = more room for solute @ temp; saturated = solute maxed out @ temp; supersaturated = more solute than maximum for given temp

4th Semester Final Exam Review Science Answer

United States Air Force Academy

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Learning as a Generative Activity Logan Fiorella, Richard E. Mayer, 2015-02-05 This book presents eight evidence based strategies that promote generative learning which enables learners to apply their knowledge to new problems

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Unforgettable W. Michael Gray, 2016-10-12 We have an uneasy relationship with the relentless deluge of information gushing out of academia and our media outlets To turn it off is escapist but to attempt to cognitively grapple with it is overwhelming In Unforgettable Enabling Deep and Durable Learning a nationally recognized master teacher gives professors and their students the means to chart a clear path through this information explosion Humans crave explanatory patterns and this book enables teachers to think deeply about their academic disciplines to find and articulate their core explanatory principles and to engage their students in a compelling way of thinking An alternative title for this book could be Why the Best College Teachers Do What They Do because the author articulates a compelling rationale that will equip faculty to create and deliver transformative courses Students in transformative courses grapple with essential questions and gain mental muscle that equips them for real world challenges *The Science Teacher*, 1968 SCC Library has 1964 cur

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