

Multiple Choice

Selecting the alternative that constitutes the statement or answers the question. What is most commonly observed with water?

1. How many moles of hydrogen gas are in a cylinder at typical conditions, 1000 L?
 a. 1
 b. 10
 c. 100
 d. 1000

2. At temperature T_1 , 1 mol of H_2 has kinetic energy E_1 . At temperature T_2 , what kinetic energy does 1 mol of H_2 have?
 a. E_1
 b. $2E_1$
 c. $4E_1$
 d. $8E_1$

3. The universal gas constant is the same for:
 a. all of the above
 b. gases with different molecular weights
 c. pure substances
 d. none of the above

4. Pure substance H_2 , 1 mol , at 100°C and 1 atm has energy E_1 . Pure substance H_2 at 100°C and 1 atm has energy E_2 . What is the difference in energy between E_1 and E_2 ?
 a. $E_1 - E_2$
 b. $E_2 - E_1$
 c. $E_1 + E_2$
 d. $E_1 E_2$

5. What is the state of the water in the following cases?
 a. liquid water
 b. solid water
 c. gaseous water
 d. none of the above

6. What is the boiling point of water at 1 atm ?
 a. 100°C
 b. 100°F
 c. 212°C
 d. 212°F

7. A substance's specific heat is $1 \text{ J/g}^\circ\text{C}$. At the same amount of heat, which substance has the highest temperature change?
 a. 1 g
 b. 10 g
 c. 100 g
 d. 1000 g

8. Pure substance H_2 , 1 mol , at 100°C and 1 atm has energy E_1 . Pure substance H_2 at 100°C and 1 atm has energy E_2 . What is the difference in energy between E_1 and E_2 ?
 a. $E_1 - E_2$
 b. $E_2 - E_1$
 c. $E_1 + E_2$
 d. $E_1 E_2$

9. At temperature T_1 , 1 mol of H_2 has kinetic energy E_1 . At temperature T_2 , what kinetic energy does 1 mol of H_2 have?
 a. E_1
 b. $2E_1$
 c. $4E_1$
 d. $8E_1$

10. What is the boiling point of water at 1 atm ?
 a. 100°C
 b. 100°F
 c. 212°C
 d. 212°F

11. The universal gas constant is the same for:
 a. all of the above
 b. gases with different molecular weights
 c. pure substances
 d. none of the above

Multiple Choice Stoichiometry Test

Richard H. Langley, John Moore



Multiple Choice Stoichiometry Test:

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web look for opportunities to contribute to the anti bullying activities these questions can be used as an icebreaker activity or to prompt small group discussions following the presentation 10 resources stopbullying.gov offers tips facts toolkits training materials and more related to bullying their inventory