

Multiple Choice:

Selecting the alternative that constitutes the statement or answers the question. What is most commonly observed with water?

1. How many moles of hydrogen gas are in a cylinder at typical conditions, 1000 L?  
 a. 1  
 b. 10  
 c. 100  
 d. 1000

2. At temperature  $T_1$ ,  $1 \text{ mol}$  of  $\text{H}_2\text{O}$  takes twice as long to evaporate as  $1 \text{ mol}$  of  $\text{H}_2$ .  
 a. 100  
 b. 10  
 c. 1  
 d. 0.1

3. The unphysical step distance in the forward run:  
 a. Half of the distance  
 b. Same as the distance  
 c. Twice the distance  
 d. Four times the distance

4. Full temperature  $T_1$ ,  $1 \text{ mol}$  of  $\text{H}_2\text{O}$  takes twice as long to evaporate as  $1 \text{ mol}$  of  $\text{H}_2$ .  
 a. 100  
 b. 10  
 c. 1  
 d. 0.1

5. What is the state of the water molecule?  
 a. Liquid  
 b. Solid  
 c. Gas  
 d. Plasma

6. What is the state of the water molecule?  
 a. Liquid  
 b. Solid  
 c. Gas  
 d. Plasma

7. A molecule's average distance in  $\text{H}_2\text{O}$  at the state of the water molecule is the same as the distance in the state of the water molecule.  
 a. 100  
 b. 10  
 c. 1  
 d. 0.1

8. Full temperature  $T_1$ ,  $1 \text{ mol}$  of  $\text{H}_2\text{O}$  takes twice as long to evaporate as  $1 \text{ mol}$  of  $\text{H}_2$ .  
 a. 100  
 b. 10  
 c. 1  
 d. 0.1

9. At temperature  $T_1$ ,  $1 \text{ mol}$  of  $\text{H}_2\text{O}$  takes twice as long to evaporate as  $1 \text{ mol}$  of  $\text{H}_2$ .  
 a. 100  
 b. 10  
 c. 1  
 d. 0.1

10. The distance between the distance of the water molecule and the distance of the water molecule.  
 a. 100  
 b. 10  
 c. 1  
 d. 0.1

11. The distance between the distance of the water molecule and the distance of the water molecule.  
 a. 100  
 b. 10  
 c. 1  
 d. 0.1

# Multiple Choice Stoichiometry Test

**Mary Millhollon, Richard H. Langley**



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